

# Imidodicarbonic acid, diethyl ester

**Inchi:** InChI=1S/C6H11NO4/c1-3-10-5(8)7-6(9)11-4-2/h3-4H2,1-2H3,(H,7,8,9)  
**InchiKey:** PQVSTLUFSYVLTO-UHFFFAOYSA-N  
**Formula:** C6H11NO4  
**SMILES:** CCOC(=O)N=C(O)OCC  
**Mol. weight [g/mol]:** 161.16  
**CAS:** 19617-44-8

## Physical Properties

Property code	Value	Unit	Source
hf	-623.99	kJ/mol	Joback Method
hvap	60.59	kJ/mol	Joback Method
log10ws	-0.74		Crippen Method
logp	1.093		Crippen Method
mcvol	120.260	ml/mol	McGowan Method
pc	3181.14	kPa	Joback Method
tb	499.20	K	NIST Webbook
tc	792.57	K	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	406.00 ± 1.00	K	2.00	NIST Webbook

## Sources

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)  
**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C19617448&Units=SI>  
**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307I>  
**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature

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