

# Butanoic acid, 2-bromo-3-methyl-

<b>Other names:</b>	Butyric acid, 2-bromo-3-methyl- «alpha»-Bromoisovaleric acid 2-Bromo-3-methylbutyric acid 2-Bromoisovaleric acid 2-Bromo-3-methylbutanoic acid NSC 167 Isovaleric acid, «alpha»-bromo-
<b>Inchi:</b>	InChI=1S/C5H9BrO2/c1-3(2)4(6)5(7)8/h3-4H,1-2H3,(H,7,8)
<b>InchiKey:</b>	UEBARDWJXBGYEJ-UHFFFAOYSA-N
<b>Formula:</b>	C5H9BrO2
<b>SMILES:</b>	CC(C)C(Br)C(=O)O
<b>Mol. weight [g/mol]:</b>	181.03
<b>CAS:</b>	565-74-2

## Physical Properties

Property code	Value	Unit	Source
gf	-265.08	kJ/mol	Joback Method
hf	-395.57	kJ/mol	Joback Method
hfus	12.63	kJ/mol	Joback Method
hvap	55.81	kJ/mol	Joback Method
log10ws	-1.32		Crippen Method
logp	1.490		Crippen Method
mcvol	106.250	ml/mol	McGowan Method
pc	4704.19	kPa	Joback Method
tb	525.13	K	Joback Method
tc	720.21	K	Joback Method
tf	286.66	K	Joback Method
vc	0.391	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	215.45	J/molxK	525.13	Joback Method
cpg	250.64	J/molxK	687.70	Joback Method

cpg	244.38	J/molxK	655.19	Joback Method
cpg	237.75	J/molxK	622.67	Joback Method
cpg	230.73	J/molxK	590.16	Joback Method
cpg	223.30	J/molxK	557.64	Joback Method
cpg	256.54	J/molxK	720.21	Joback Method
dvisc	0.0001851	Paxs	525.13	Joback Method
dvisc	0.0003004	Paxs	485.38	Joback Method
dvisc	0.0005317	Paxs	445.64	Joback Method
dvisc	0.0010525	Paxs	405.89	Joback Method
dvisc	0.0024162	Paxs	366.15	Joback Method
dvisc	0.0067911	Paxs	326.41	Joback Method
dvisc	0.0254209	Paxs	286.66	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	398.20	K	2.70	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C565742&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C565742&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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