

«alpha»,«alpha»-Dichloromethyl methyl ether

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| Other names: | (Dichloromethyl)methyl ether CH ₃ OCHCl ₂ 1,1-Dichloromethylmethyl ether Methane, dichloromethoxy- Dichloromethoxymethane Ether, dichloromethyl methyl «alpha»,«alpha»-Dichloromethyl ether Methyl dichloromethyl ether |
| Inchi: | InChI=1S/C2H4Cl2O/c1-5-2(3)4/h2H,1H3 |
| InchiKey: | GRTGGSXWHGKRSB-UHFFFAOYSA-N |
| Formula: | C ₂ H ₄ Cl ₂ O |
| SMILES: | COC(Cl)Cl |
| Mol. weight [g/mol]: | 114.96 |
| CAS: | 4885-02-3 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -165.34 | kJ/mol | Joback Method |
| hf | -253.59 | kJ/mol | Joback Method |
| hfus | 7.00 | kJ/mol | Joback Method |
| hvap | 30.84 | kJ/mol | Joback Method |
| log10ws | -1.16 | | Crippen Method |
| logp | 1.394 | | Crippen Method |
| mcvol | 69.390 | ml/mol | McGowan Method |
| pc | 4486.22 | kPa | Joback Method |
| tb | 342.00 | K | Joback Method |
| tc | 530.03 | K | Joback Method |
| tf | 179.37 | K | Joback Method |
| vc | 0.258 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-------|---------|-----------------|---------------|
| cpg | 98.92 | J/mol×K | 342.00 | Joback Method |

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|-------|-----------|---------|--------|---------------|
| cpg | 117.99 | J/molxK | 498.69 | Joback Method |
| cpg | 114.38 | J/molxK | 467.35 | Joback Method |
| cpg | 110.66 | J/molxK | 436.02 | Joback Method |
| cpg | 106.84 | J/molxK | 404.68 | Joback Method |
| cpg | 102.92 | J/molxK | 373.34 | Joback Method |
| cpg | 121.50 | J/molxK | 530.03 | Joback Method |
| dvisc | 0.0003110 | Paxs | 342.00 | Joback Method |
| dvisc | 0.0004002 | Paxs | 314.89 | Joback Method |
| dvisc | 0.0005399 | Paxs | 287.79 | Joback Method |
| dvisc | 0.0007752 | Paxs | 260.69 | Joback Method |
| dvisc | 0.0012106 | Paxs | 233.58 | Joback Method |
| dvisc | 0.0021252 | Paxs | 206.47 | Joback Method |
| dvisc | 0.0044225 | Paxs | 179.37 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C4885023&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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