

# 1,4-Cyclohexanediamine, cis-

<b>Other names:</b>	cis-1,4-Diaminocyclohexane cis-1,4-cyclohexanediamine
<b>Inchi:</b>	InChI=1S/C6H14N2/c7-5-1-2-6(8)4-3-5/h5-6H,1-4,7-8H2/t5-,6+
<b>InchiKey:</b>	VKIRRGRTJUUZHS-OLQVQODUSA-N
<b>Formula:</b>	C6H14N2
<b>SMILES:</b>	NC1CCC(N)CC1
<b>Mol. weight [g/mol]:</b>	114.19
<b>CAS:</b>	15827-56-2

## Physical Properties

Property code	Value	Unit	Source
gf	149.28	kJ/mol	Joback Method
hf	-65.61	kJ/mol	Joback Method
hfus	14.60	kJ/mol	Joback Method
hvap	50.35	kJ/mol	Joback Method
log10ws	-1.32		Crippen Method
logp	0.215		Crippen Method
mvol	104.500	ml/mol	McGowan Method
pc	4397.41	kPa	Joback Method
tb	496.62	K	Joback Method
tc	730.03	K	Joback Method
tf	327.04	K	Joback Method
vc	0.361	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	321.72	J/mol×K	691.13	Joback Method
cpg	250.11	J/mol×K	496.62	Joback Method
cpg	266.28	J/mol×K	535.52	Joback Method
cpg	281.50	J/mol×K	574.42	Joback Method
cpg	295.80	J/mol×K	613.32	Joback Method
cpg	309.20	J/mol×K	652.22	Joback Method
cpg	333.38	J/mol×K	730.03	Joback Method

hvapt	62.20	kJ/mol	298.15	Enthalpy of sublimation/vaporization of trans-cyclohexyl-1,4-diamine and cis-cyclohexyl-1,2-diamine
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## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Enthalpy of sublimation/vaporization of trans-cyclohexyl-1,4-diamine and cis-cyclohexyl-1,2-diamine:</b>	<a href="https://www.doi.org/10.1016/j.jct.2007.06.015">https://www.doi.org/10.1016/j.jct.2007.06.015</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15827562&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15827562&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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