

# Triethylene glycol, monoallyl ether, acetate

**Inchi:** InChI=1S/C11H20O5/c1-3-4-13-5-6-14-7-8-15-9-10-16-11(2)12/h3H,1,4-10H2,2H3  
**InchiKey:** MHLDUOGECSISSO-UHFFFAOYSA-N  
**Formula:** C11H20O5  
**SMILES:** C=CCOCCOCCOCCOC(C)=O  
**Mol. weight [g/mol]:** 232.27

## Physical Properties

Property code	Value	Unit	Source
gf	-419.34	kJ/mol	Joback Method
hf	-786.40	kJ/mol	Joback Method
hfus	29.32	kJ/mol	Joback Method
hvap	55.80	kJ/mol	Joback Method
log10ws	-0.41		Crippen Method
logp	0.785		Crippen Method
mcvol	186.600	ml/mol	McGowan Method
pc	1998.33	kPa	Joback Method
rinpol	1527.00		NIST Webbook
rinpol	1531.00		NIST Webbook
rinpol	1527.00		NIST Webbook
rinpol	1529.00		NIST Webbook
rinpol	1524.00		NIST Webbook
rinpol	1530.00		NIST Webbook
rinpol	1524.00		NIST Webbook
rinpol	1526.00		NIST Webbook
rinpol	1526.00		NIST Webbook
rinpol	1522.00		NIST Webbook
rinpol	1529.00		NIST Webbook
rinpol	1526.00		NIST Webbook
rinpol	1529.00		NIST Webbook
rinpol	1530.00		NIST Webbook
tb	591.31	K	Joback Method
tc	763.89	K	Joback Method
tf	350.82	K	Joback Method
vc	0.711	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	475.59	J/molxK	591.31	Joback Method
cpg	489.53	J/molxK	620.07	Joback Method
cpg	502.95	J/molxK	648.84	Joback Method
cpg	515.87	J/molxK	677.60	Joback Method
cpg	528.25	J/molxK	706.36	Joback Method
cpg	540.09	J/molxK	735.12	Joback Method
cpg	551.38	J/molxK	763.89	Joback Method
dvisc	0.0011186	Paxs	350.82	Joback Method
dvisc	0.0006206	Paxs	390.90	Joback Method
dvisc	0.0003842	Paxs	430.98	Joback Method
dvisc	0.0002581	Paxs	471.06	Joback Method
dvisc	0.0001845	Paxs	511.15	Joback Method
dvisc	0.0001385	Paxs	551.23	Joback Method
dvisc	0.0001081	Paxs	591.31	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R152300&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R152300&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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