

# Carbonic acid, (1R)-(-)-menthyl nonyl ester

**Inchi:** InChI=1S/C20H38O3/c1-5-6-7-8-9-10-11-14-22-20(21)23-19-15-18(16(2)3)13-12-17(19)  
**InchiKey:** DLMWJBSSSYXANL-UHFFFAOYSA-N  
**Formula:** C20H38O3  
**SMILES:** CCCCCCCCCOC(=O)OC1CC(C(C)C)CCC1C  
**Mol. weight [g/mol]:** 326.51

## Physical Properties

Property code	Value	Unit	Source
gf	-214.81	kJ/mol	Joback Method
hf	-824.79	kJ/mol	Joback Method
hfus	41.98	kJ/mol	Joback Method
hvap	71.10	kJ/mol	Joback Method
log10ws	-6.40		Crippen Method
logp	6.351		Crippen Method
mvol	295.110	ml/mol	McGowan Method
pc	1139.80	kPa	Joback Method
rinpol	2179.00		NIST Webbook
rinpol	2179.00		NIST Webbook
tb	765.48	K	Joback Method
tc	954.07	K	Joback Method
tf	393.45	K	Joback Method
vc	1.123	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	944.61	J/molxK	765.48	Joback Method
cpg	966.32	J/molxK	796.91	Joback Method
cpg	986.75	J/molxK	828.34	Joback Method
cpg	1005.90	J/molxK	859.77	Joback Method
cpg	1023.79	J/molxK	891.20	Joback Method
cpg	1040.42	J/molxK	922.64	Joback Method
cpg	1055.81	J/molxK	954.07	Joback Method
dvisc	0.0015294	Paxs	393.45	Joback Method

dvisc	0.0006781	Paxs	455.46	Joback Method
dvisc	0.0003653	Paxs	517.46	Joback Method
dvisc	0.0002247	Paxs	579.47	Joback Method
dvisc	0.0001518	Paxs	641.47	Joback Method
dvisc	0.0001099	Paxs	703.48	Joback Method
dvisc	0.0000838	Paxs	765.48	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U392438&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U392438&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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