

# Isophthalic acid, heptyl 1-phenylpropyl ester

<b>Inchi:</b>	InChI=1S/C24H30O4/c1-3-5-6-7-11-17-27-23(25)20-15-12-16-21(18-20)24(26)28-22(4-2
<b>InchiKey:</b>	BCZMLXWJMITYDS-UHFFFAOYSA-N
<b>Formula:</b>	C24H30O4
<b>SMILES:</b>	CCCCCCCOC(=O)c1cccc(C(=O)OC(CC)c2ccccc2)c1
<b>Mol. weight [g/mol]:</b>	382.49

## Physical Properties

Property code	Value	Unit	Source
gf	-103.89	kJ/mol	Joback Method
hf	-571.98	kJ/mol	Joback Method
hfus	47.66	kJ/mol	Joback Method
hvap	92.16	kJ/mol	Joback Method
log10ws	-7.38		Crippen Method
logp	6.122		Crippen Method
mvol	316.380	ml/mol	McGowan Method
pc	1298.60	kPa	Joback Method
rinpol	2951.00		NIST Webbook
rinpol	2951.00		NIST Webbook
tb	959.00	K	Joback Method
tc	1183.19	K	Joback Method
tf	554.92	K	Joback Method
vc	1.206	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1017.17	J/molxK	959.00	Joback Method
cpg	1031.66	J/molxK	996.37	Joback Method
cpg	1044.74	J/molxK	1033.73	Joback Method
cpg	1056.47	J/molxK	1071.10	Joback Method
cpg	1066.89	J/molxK	1108.46	Joback Method
cpg	1076.05	J/molxK	1145.83	Joback Method
cpg	1084.02	J/molxK	1183.19	Joback Method
dvisc	0.0003715	Paxs	554.92	Joback Method

dvisc	0.0001929	Paxs	622.27	Joback Method
dvisc	0.0001139	Paxs	689.61	Joback Method
dvisc	0.0000738	Paxs	756.96	Joback Method
dvisc	0.0000514	Paxs	824.31	Joback Method
dvisc	0.0000377	Paxs	891.65	Joback Method
dvisc	0.0000290	Paxs	959.00	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344553&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344553&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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