

# 3,4-Methylenedioxypropiphenone

<b>Other names:</b>	1-Propanone, 1-(1,3-benzodioxol-5-yl)- 1-(1,3-Benzodioxol-5-yl)-1-propanone 3',4'-Methylenedioxypropiphenone NSC 29484 1-(1,3-benzodioxol-5-yl)propan-1-one
<b>Inchi:</b>	InChI=1S/C10H10O3/c1-2-8(11)7-3-4-9-10(5-7)13-6-12-9/h3-5H,2,6H2,1H3
<b>InchiKey:</b>	RVBJGSPBFIUTTR-UHFFFAOYSA-N
<b>Formula:</b>	C10H10O3
<b>SMILES:</b>	CCC(=O)c1ccc2c(c1)OCO2
<b>Mol. weight [g/mol]:</b>	178.18
<b>CAS:</b>	28281-49-4

## Physical Properties

Property code	Value	Unit	Source
gf	-106.23	kJ/mol	Joback Method
hf	-319.58	kJ/mol	Joback Method
h <sub>fus</sub>	29.54	kJ/mol	Joback Method
h <sub>vap</sub>	57.44	kJ/mol	Joback Method
log <sub>10</sub> ws	-2.76		Crippen Method
logp	2.008		Crippen Method
m <sub>cvol</sub>	130.450	ml/mol	McGowan Method
pc	3589.94	kPa	Joback Method
tb	584.02	K	Joback Method
tc	813.44	K	Joback Method
tf	379.17	K	Joback Method
vc	0.493	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
c <sub>pg</sub>	315.82	J/mol×K	584.02	Joback Method
c <sub>pg</sub>	328.09	J/mol×K	622.26	Joback Method
c <sub>pg</sub>	339.48	J/mol×K	660.49	Joback Method
c <sub>pg</sub>	350.06	J/mol×K	698.73	Joback Method

cpg	359.87	J/molxK	736.97	Joback Method
cpg	368.99	J/molxK	775.21	Joback Method
cpg	377.47	J/molxK	813.44	Joback Method
dvisc	0.0022370	Paxs	379.17	Joback Method
dvisc	0.0015736	Paxs	413.31	Joback Method
dvisc	0.0011680	Paxs	447.45	Joback Method
dvisc	0.0009043	Paxs	481.60	Joback Method
dvisc	0.0007243	Paxs	515.74	Joback Method
dvisc	0.0005964	Paxs	549.88	Joback Method
dvisc	0.0005023	Paxs	584.02	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	439.50 ± 1.50	K	2.70	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C28281494&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C28281494&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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