

# 2-Hydroxy-iso-butyrophenone

<b>Other names:</b>	1-Propanone, 2-hydroxy-2-methyl-1-phenyl- 1-Phenyl-2-hydroxy-2-methylpropan-1-one 2-Hydroxy-2-methylpropiophenone 2-Hydroxy-2-methyl-1-phenyl-1-propanone 2-Hydroxy-2-methyl-1-phenylpropan-1-one Darocure-173 (2-Hydroxy-2-methyl-1-phenyl-propanone)
<b>Inchi:</b>	InChI=1S/C10H12O2/c1-10(2,12)9(11)8-6-4-3-5-7-8/h3-7,12H,1-2H3
<b>InchiKey:</b>	XMLYCEVDHLAQEL-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O2
<b>SMILES:</b>	CC(C)(O)C(=O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	164.20
<b>CAS:</b>	7473-98-5

## Physical Properties

Property code	Value	Unit	Source
gf	-117.17	kJ/mol	Joback Method
hf	-286.76	kJ/mol	Joback Method
hfus	13.97	kJ/mol	Joback Method
hvap	62.26	kJ/mol	Joback Method
log10ws	-2.34		Crippen Method
logp	1.640		Crippen Method
mcvol	135.440	ml/mol	McGowan Method
pc	3572.80	kPa	Joback Method
rinpol	1278.00		NIST Webbook
tb	597.70	K	Joback Method
tc	809.57	K	Joback Method
tf	342.05	K	Joback Method
vc	0.501	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	331.24	J/molxK	597.70	Joback Method
cpg	343.11	J/molxK	633.01	Joback Method

cpg	354.13	J/mol×K	668.32	Joback Method
cpg	364.35	J/mol×K	703.63	Joback Method
cpg	373.81	J/mol×K	738.95	Joback Method
cpg	382.59	J/mol×K	774.26	Joback Method
cpg	390.72	J/mol×K	809.57	Joback Method
dvisc	0.0065486	Paxs	342.05	Joback Method
dvisc	0.0021138	Paxs	384.66	Joback Method
dvisc	0.0008549	Paxs	427.27	Joback Method
dvisc	0.0004074	Paxs	469.88	Joback Method
dvisc	0.0002197	Paxs	512.48	Joback Method
dvisc	0.0001302	Paxs	555.09	Joback Method
dvisc	0.0000832	Paxs	597.70	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7473985&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7473985&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/85-851-3/2-Hydroxy-iso-butyrophenone.pdf>

Generated by Cheméo on 2024-07-18 11:22:11.049870301 +0000 UTC m=+527946.245840673.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.