

2-(3',4'-Methylenedioxy)phenyl-4-keto-1,3-benzoth

Inchi:	InChI=1S/C15H10O4S/c16-14-10-3-1-2-4-13(10)20-15(19-14)9-5-6-11-12(7-9)18-8-17-1
InchiKey:	FFXRKNMJMJPRBJ-UHFFFAOYSA-N
Formula:	C15H10O4S
SMILES:	O=C1OC(c2ccc3c(c2)OCO3)Sc2ccccc21
Mol. weight [g/mol]:	286.30
CAS:	97192-88-6

Physical Properties

Property code	Value	Unit	Source
gf	47.37	kJ/mol	Joback Method
hf	-242.94	kJ/mol	Joback Method
hfus	41.72	kJ/mol	Joback Method
hvap	79.42	kJ/mol	Joback Method
log10ws	-4.71		Crippen Method
logp	3.377		Crippen Method
mcvol	188.500	ml/mol	McGowan Method
pc	3376.28	kPa	Joback Method
tb	829.82	K	Joback Method
tc	1116.87	K	Joback Method
tf	617.19	K	Joback Method
vc	0.682	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	543.41	J/molxK	829.82	Joback Method
cpg	556.84	J/molxK	877.66	Joback Method
cpg	568.98	J/molxK	925.50	Joback Method
cpg	579.97	J/molxK	973.35	Joback Method
cpg	589.94	J/molxK	1021.19	Joback Method
cpg	599.03	J/molxK	1069.03	Joback Method
cpg	607.38	J/molxK	1116.87	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C97192886&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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