

# 3-Phenylpropionic acid, phenyl ester

<b>Inchi:</b>	InChI=1S/C15H14O2/c16-15(17-14-9-5-2-6-10-14)12-11-13-7-3-1-4-8-13/h1-10H,11-12H
<b>InchiKey:</b>	VIGODTIMSIHKSD-UHFFFAOYSA-N
<b>Formula:</b>	C15H14O2
<b>SMILES:</b>	O=C(CCc1ccccc1)Oc1ccccc1
<b>Mol. weight [g/mol]:</b>	226.27

## Physical Properties

Property code	Value	Unit	Source
gf	66.32	kJ/mol	Joback Method
hf	-124.67	kJ/mol	Joback Method
hfus	25.48	kJ/mol	Joback Method
hvap	62.69	kJ/mol	Joback Method
log10ws	-3.82		Crippen Method
logp	3.225		Crippen Method
mcvol	182.130	ml/mol	McGowan Method
pc	2629.85	kPa	Joback Method
rinsol	1801.00		NIST Webbook
tb	672.25	K	Joback Method
tc	909.92	K	Joback Method
tf	383.81	K	Joback Method
vc	0.683	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	466.75	J/molxK	672.25	Joback Method
cpg	482.53	J/molxK	711.86	Joback Method
cpg	497.08	J/molxK	751.47	Joback Method
cpg	510.44	J/molxK	791.09	Joback Method
cpg	522.68	J/molxK	830.70	Joback Method
cpg	533.85	J/molxK	870.31	Joback Method
cpg	544.01	J/molxK	909.92	Joback Method
dvisc	0.0015219	Paxs	383.81	Joback Method
dvisc	0.0008139	Paxs	431.88	Joback Method

dvisc	0.0004934	Paxs	479.96	Joback Method
dvisc	0.0003277	Paxs	528.03	Joback Method
dvisc	0.0002330	Paxs	576.10	Joback Method
dvisc	0.0001746	Paxs	624.18	Joback Method
dvisc	0.0001364	Paxs	672.25	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U299021&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U299021&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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