

# Carbonic acid, (1R)-(-)-menthyl decyl ester

**Inchi:** InChI=1S/C21H40O3/c1-5-6-7-8-9-10-11-12-15-23-21(22)24-20-16-19(17(2)3)14-13-18(2)  
**InchiKey:** SHGKNWDDQRWFPQ-UHFFFAOYSA-N  
**Formula:** C21H40O3  
**SMILES:** CCCCCCCCCCOC(=O)OC1CC(C(C)C)CCC1C  
**Mol. weight [g/mol]:** 340.54

## Physical Properties

Property code	Value	Unit	Source
gf	-206.39	kJ/mol	Joback Method
hf	-845.43	kJ/mol	Joback Method
hfus	44.57	kJ/mol	Joback Method
hvap	73.33	kJ/mol	Joback Method
log10ws	-6.82		Crippen Method
logp	6.741		Crippen Method
mcvol	309.200	ml/mol	McGowan Method
pc	1069.36	kPa	Joback Method
rinsol	2285.00		NIST Webbook
rinsol	2285.00		NIST Webbook
tb	788.36	K	Joback Method
tc	977.48	K	Joback Method
tf	404.72	K	Joback Method
vc	1.179	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1006.55	J/molxK	788.36	Joback Method
cpg	1102.31	J/molxK	945.96	Joback Method
cpg	1085.78	J/molxK	914.44	Joback Method
cpg	1067.95	J/molxK	882.92	Joback Method
cpg	1048.81	J/molxK	851.40	Joback Method
cpg	1028.35	J/molxK	819.88	Joback Method
cpg	1117.57	J/molxK	977.48	Joback Method
dvisc	0.0000739	Paxs	788.36	Joback Method

dvisc	0.0000971	Paxs	724.42	Joback Method
dvisc	0.0001347	Paxs	660.48	Joback Method
dvisc	0.0002002	Paxs	596.54	Joback Method
dvisc	0.0003275	Paxs	532.60	Joback Method
dvisc	0.0006127	Paxs	468.66	Joback Method
dvisc	0.0013969	Paxs	404.72	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U392439&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U392439&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/84-772-2/Carbonic-acid-1R-menthyl-decyl-ester.pdf>

Generated by Cheméo on 2024-04-27 16:23:50.242120398 +0000 UTC m=+16524279.162697719.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.