

# 2-Hydroxymethyl-1,3-dioxane

<b>Inchi:</b>	InChI=1S/C5H10O3/c6-4-5-7-2-1-3-8-5/h5-6H,1-4H2
<b>InchiKey:</b>	IEBHUKYDOBADIB-UHFFFAOYSA-N
<b>Formula:</b>	C5H10O3
<b>SMILES:</b>	OCC1OCCCO1
<b>Mol. weight [g/mol]:</b>	118.13
<b>CAS:</b>	39239-93-5

## Physical Properties

Property code	Value	Unit	Source
chl	-2773.60 ± 4.90	kJ/mol	NIST Webbook
gf	-293.39	kJ/mol	Joback Method
hf	-554.00 ± 12.00	kJ/mol	NIST Webbook
hfl	-623.00 ± 10.00	kJ/mol	NIST Webbook
hfus	20.59	kJ/mol	Joback Method
hvap	69.00 ± 2.00	kJ/mol	NIST Webbook
hvap	69.00	kJ/mol	NIST Webbook
log10ws	0.14		Crippen Method
logp	-0.258		Crippen Method
mcvol	88.060	ml/mol	McGowan Method
pc	4966.33	kPa	Joback Method
tb	479.43	K	Joback Method
tc	675.81	K	Joback Method
tf	267.45	K	Joback Method
vc	0.309	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	203.46	J/mol×K	479.43	Joback Method
cpg	253.26	J/mol×K	643.08	Joback Method
cpg	244.35	J/mol×K	610.35	Joback Method
cpg	234.93	J/mol×K	577.62	Joback Method
cpg	224.98	J/mol×K	544.89	Joback Method
cpg	214.49	J/mol×K	512.16	Joback Method

cpg	261.67	J/mol×K	675.81	Joback Method
dvisc	0.0002452	Paxs	479.43	Joback Method
dvisc	0.0004087	Paxs	444.10	Joback Method
dvisc	0.0007442	Paxs	408.77	Joback Method
dvisc	0.0015175	Paxs	373.44	Joback Method
dvisc	0.0035916	Paxs	338.11	Joback Method
dvisc	0.0103930	Paxs	302.78	Joback Method
dvisc	0.0398209	Paxs	267.45	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C39239935&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C39239935&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/83-882-1/2-Hydroxymethyl-1-3-dioxane.pdf>

Generated by Cheméo on 2024-04-23 11:02:08.109379097 +0000 UTC m=+16159377.029956412.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.