

# Etilefrine

<b>Other names:</b>	Benzenemethanol, «alpha»-[(ethylamino)methyl]-3-hydroxy- Benzyl alcohol, «alpha»-((ethylamino)methyl)-m-hydroxy- Effortil Ethyladrianol «alpha»-((Ethylamino)methyl)-m-hydroxybenzyl alcohol Ethyl noradrianol N-Ethylnorphenylephrine m-Hydroxyphenylethanoethylamine 1-(3'-Hydroxyphenyl)-2-ethylaminoethanol Ethylphenylephrine Etiladrianol Etilefrin (.+/-)-Etilefrin (.+/-)-Etilefrine 1-(3-Hydroxyphenyl)-2-(ethylamino)ethanol Racemic etilefrine S 40032 S 40032-7
<b>Inchi:</b>	InChI=1S/C10H15NO2/c1-2-11-7-10(13)8-4-3-5-9(12)6-8/h3-6,10-13H,2,7H2,1H3
<b>InchiKey:</b>	SQVI AVUSQAWMKL-UHFFFAOYSA-N
<b>Formula:</b>	C10H15NO2
<b>SMILES:</b>	CCNCC(O)c1cccc(O)c1
<b>Mol. weight [g/mol]:</b>	181.23
<b>CAS:</b>	709-55-7

## Physical Properties

Property code	Value	Unit	Source
gf	-58.76	kJ/mol	Joback Method
hf	-294.55	kJ/mol	Joback Method
hfus	27.14	kJ/mol	Joback Method
hvap	75.87	kJ/mol	Joback Method
log10ws	-1.57		Crippen Method
logp	1.035		Crippen Method
mcvol	149.720	ml/mol	McGowan Method
pc	3980.54	kPa	Joback Method
rinpol	1685.00		NIST Webbook
rinpol	1685.00		NIST Webbook

tb	677.41	K	Joback Method
tc	884.24	K	Joback Method
tf	439.08	K	Joback Method
vc	0.501	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	410.00	J/mol×K	677.41	Joback Method
cpg	421.16	J/mol×K	711.88	Joback Method
cpg	431.65	J/mol×K	746.35	Joback Method
cpg	441.55	J/mol×K	780.83	Joback Method
cpg	450.93	J/mol×K	815.30	Joback Method
cpg	459.86	J/mol×K	849.77	Joback Method
cpg	468.40	J/mol×K	884.24	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C709557&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C709557&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/83-380-8/Etilefrine.pdf>

Generated by Cheméo on 2024-05-08 06:39:31.059480753 +0000 UTC m=+17439619.980058086.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.