

# Butanamide, N-hexyl-3-methyl

Inchi:	lnChI=1S/C11H23NO/c1-4-5-6-7-8-12-11(13)9-10(2)3/h10H,4-9H2,1-3H3,(H,12,13)
InchiKey:	PZJQRNSSYCNVMU-UHFFFAOYSA-N
Formula:	C11H23NO
SMILES:	CCCCCCN=C(O)CC(C)C
Mol. weight [g/mol]:	185.31

## Physical Properties

Property code	Value	Unit	Source
hf	-355.45	kJ/mol	Joback Method
hvap	59.77	kJ/mol	Joback Method
log10ws	-3.17		Crippen Method
logp	3.569		Crippen Method
mcvol	177.400	ml/mol	McGowan Method
pc	1933.83	kPa	Joback Method
rinpol	1469.00		NIST Webbook
tb	619.38	K	Joback Method
tc	796.97	K	Joback Method

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R50494&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R50494&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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