

# 11-Octadecenoic acid, (E)-

<b>Other names:</b>	(E)-octadec-11-enoic acid Vaccenic acid trans-11-octadecenoic acid trans-Vaccenic acid
<b>Inchi:</b>	InChI=1S/C18H34O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18(19)20/h8-9H,2-7,1
<b>InchiKey:</b>	QXJSBBXBKPUZAA-CMDGGGOBGSA-N
<b>Formula:</b>	C18H34O2
<b>SMILES:</b>	CCCCCCCC=CCCCCCCCC(=O)O
<b>Mol. weight [g/mol]:</b>	282.46
<b>CAS:</b>	693-72-1

## Physical Properties

Property code	Value	Unit	Source
gf	-84.84	kJ/mol	Joback Method
hf	-562.44	kJ/mol	Joback Method
hfus	48.26	kJ/mol	Joback Method
hvap	79.05	kJ/mol	Joback Method
log10ws	-6.31		Crippen Method
logp	6.109		Crippen Method
mcvol	267.620	ml/mol	McGowan Method
pc	1332.96	kPa	Joback Method
tb	761.45	K	Joback Method
tc	937.21	K	Joback Method
tf	316.65 ± 1.00	K	NIST Webbook
vc	1.048	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	811.35	J/mol×K	761.45	Joback Method
cpg	827.83	J/mol×K	790.74	Joback Method
cpg	843.52	J/mol×K	820.04	Joback Method
cpg	858.48	J/mol×K	849.33	Joback Method
cpg	872.72	J/mol×K	878.62	Joback Method

cpg	886.31	J/mol×K	907.92	Joback Method
cpg	899.26	J/mol×K	937.21	Joback Method
dvisc	0.0006419	Paxs	458.82	Joback Method
dvisc	0.0023727	Paxs	398.29	Joback Method
dvisc	0.0002355	Paxs	519.34	Joback Method
dvisc	0.0001065	Paxs	579.87	Joback Method
dvisc	0.0000560	Paxs	640.40	Joback Method
dvisc	0.0000329	Paxs	700.92	Joback Method
dvisc	0.0000210	Paxs	761.45	Joback Method
hfust	58.49	kJ/mol	317.00	NIST Webbook
hfust	55.62	kJ/mol	318.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52924e+01
Coeff. B	-5.81882e+03
Coeff. C	-1.25923e+02
Temperature range (K), min.	514.15
Temperature range (K), max.	709.15

## Sources

**The Yaws Handbook of Vapor**

**Pressure:**

**Crippen Method:**

**Crippen Method:**

**Joback Method:**

**McGowan Method:**

**NIST Webbook:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci990307i>

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

<http://link.springer.com/article/10.1007/BF02311772>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C693721&Units=SI>

## Legend

**cpg:** Ideal gas heat capacity

**dvisc:** Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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