

# Succinic acid, dodec-2-en-1-yl 2-methoxyphenyl ester

**Inchi:** InChI=1S/C23H34O5/c1-3-4-5-6-7-8-9-10-11-14-19-27-22(24)17-18-23(25)28-21-16-13-12  
**InchiKey:** MUBZVTPSYBOHEO-SDNWHVVSQSA-N  
**Formula:** C23H34O5  
**SMILES:** CCCCCCCCC=CCOC(=O)CCC(=O)Oc1ccccc1OC  
**Mol. weight [g/mol]:** 390.51

## Physical Properties

Property code	Value	Unit	Source
gf	-247.06	kJ/mol	Joback Method
hf	-797.59	kJ/mol	Joback Method
hfus	55.94	kJ/mol	Joback Method
hvap	90.41	kJ/mol	Joback Method
log10ws	-6.48		Crippen Method
logp	5.621		Crippen Method
mvol	327.620	ml/mol	McGowan Method
pc	1123.81	kPa	Joback Method
rinpol	2901.00		NIST Webbook
rinpol	2901.00		NIST Webbook
tb	936.46	K	Joback Method
tc	1148.00	K	Joback Method
tf	549.38	K	Joback Method
vc	1.262	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1059.82	J/molxK	936.46	Joback Method
cpg	1075.61	J/molxK	971.72	Joback Method
cpg	1090.07	J/molxK	1006.97	Joback Method
cpg	1103.24	J/molxK	1042.23	Joback Method
cpg	1115.16	J/molxK	1077.49	Joback Method
cpg	1125.86	J/molxK	1112.74	Joback Method
cpg	1135.37	J/molxK	1148.00	Joback Method
dvisc	0.0002765	Paxs	549.38	Joback Method

dvisc	0.0001473	Paxs	613.89	Joback Method
dvisc	0.0000884	Paxs	678.41	Joback Method
dvisc	0.0000580	Paxs	742.92	Joback Method
dvisc	0.0000407	Paxs	807.43	Joback Method
dvisc	0.0000301	Paxs	871.95	Joback Method
dvisc	0.0000232	Paxs	936.46	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389719&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389719&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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