

# Butanoic acid, 3-hydroxy-2,2-dimethyl-, ethyl ester

Other names:	Ethyl 3-hydroxy-2,2-dimethylbutanoate
Inchi:	InChI=1S/C8H16O3/c1-5-11-7(10)8(3,4)6(2)9/h6,9H,5H2,1-4H3
InchiKey:	DPZVLAWBYJYWFX-UHFFFAOYSA-N
Formula:	C8H16O3
SMILES:	CCOC(=O)C(C)(C)C(C)O
Mol. weight [g/mol]:	160.21
CAS:	7505-94-4

## Physical Properties

Property code	Value	Unit	Source
gf	-353.86	kJ/mol	Joback Method
hf	-619.51	kJ/mol	Joback Method
hfus	12.41	kJ/mol	Joback Method
hvap	57.55	kJ/mol	Joback Method
log10ws	-1.17		Crippen Method
logp	0.956		Crippen Method
mcvol	136.890	ml/mol	McGowan Method
pc	3015.64	kPa	Joback Method
ripol	1699.00		NIST Webbook
ripol	1699.00		NIST Webbook
tb	547.24	K	Joback Method
tc	728.02	K	Joback Method
tf	300.32	K	Joback Method
vc	0.509	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	337.14	J/molxK	547.24	Joback Method
cpg	348.80	J/molxK	577.37	Joback Method
cpg	359.89	J/molxK	607.50	Joback Method
cpg	370.42	J/molxK	637.63	Joback Method
cpg	380.43	J/molxK	667.76	Joback Method
cpg	389.92	J/molxK	697.89	Joback Method

cpg	398.91	J/mol×K	728.02	Joback Method
dvisc	0.0169327	Paxs	300.32	Joback Method
dvisc	0.0042752	Paxs	341.47	Joback Method
dvisc	0.0014513	Paxs	382.63	Joback Method
dvisc	0.0006077	Paxs	423.78	Joback Method
dvisc	0.0002969	Paxs	464.93	Joback Method
dvisc	0.0001629	Paxs	506.09	Joback Method
dvisc	0.0000979	Paxs	547.24	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7505944&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7505944&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/81-347-7/Butanoic-acid-3-hydroxy-2-2-dimethyl-ethyl-ester.pdf>

Generated by Cheméo on 2024-05-03 20:52:15.9076623 +0000 UTC m=+17058784.828239616.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.