

# Benzene, 1-fluoro-3-iodo-

<b>Other names:</b>	1,3-fluoriodobenzene 1-fluoro-3-iodobenzene 3-Fluoriodobenzene 3-Iodofluorobenzene m-Fluorobenzene m-Fluoriodobenzene m-Iodofluorobenzene
<b>Inchi:</b>	InChI=1S/C6H4FI/c7-5-2-1-3-6(8)4-5/h1-4H
<b>InchiKey:</b>	VSKSBSORLCDRHS-UHFFFAOYSA-N
<b>Formula:</b>	C6H4FI
<b>SMILES:</b>	Fc1cccc(I)c1
<b>Mol. weight [g/mol]:</b>	222.00
<b>CAS:</b>	1121-86-4

## Physical Properties

Property code	Value	Unit	Source
gf	-34.27	kJ/mol	Joback Method
hf	-61.35	kJ/mol	Joback Method
hfus	12.43	kJ/mol	Joback Method
hvap	40.44	kJ/mol	Joback Method
log10ws	-2.95		Crippen Method
logp	2.430		Crippen Method
mcvol	99.230	ml/mol	McGowan Method
pc	4189.32	kPa	Joback Method
tb	460.75	K	Joback Method
tc	704.26	K	Joback Method
tf	254.97	K	Joback Method
vc	0.369	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	152.63	J/mol×K	460.75	Joback Method
cpg	161.13	J/mol×K	501.33	Joback Method

cpg	168.95	J/mol×K	541.92	Joback Method
cpg	176.13	J/mol×K	582.50	Joback Method
cpg	182.73	J/mol×K	623.09	Joback Method
cpg	188.79	J/mol×K	663.67	Joback Method
cpg	194.34	J/mol×K	704.26	Joback Method
hvapt	48.60	kJ/mol	298.15	Vaporization enthalpies of a series of the halogen-substituted fluorobenzenes

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	350.70	K	2.50	NIST Webbook
tbrp	350.50 ± 0.50	K	2.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1121864&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1121864&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Vaporization enthalpies of a series of the halogen-substituted fluorobenzenes:</b>	<a href="https://www.doi.org/10.1016/j.fluid.2014.12.023">https://www.doi.org/10.1016/j.fluid.2014.12.023</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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