

# 1,2-Ethanediol, diformate

<b>Other names:</b>	Ethylene glycol, diformate Ethylene formate Glycol, diformate 1,2-Diformyloxyethane Formic acid, ethylene ester Ethanediol diformate ethylene diformate
<b>Inchi:</b>	InChI=1S/C4H6O4/c5-3-7-1-2-8-4-6/h3-4H,1-2H2
<b>InchiKey:</b>	IKCQWKJZLSDDSS-UHFFFAOYSA-N
<b>Formula:</b>	C4H6O4
<b>SMILES:</b>	O=COCCOC=O
<b>Mol. weight [g/mol]:</b>	118.09
<b>CAS:</b>	629-15-2

## Physical Properties

Property code	Value	Unit	Source
gf	-426.24	kJ/mol	Joback Method
hf	-561.49	kJ/mol	Joback Method
hfus	13.07	kJ/mol	Joback Method
hvap	42.76	kJ/mol	Joback Method
log10ws	0.78		Crippen Method
logp	-0.668		Crippen Method
mvol	82.100	ml/mol	McGowan Method
pc	4516.42	kPa	Joback Method
tb	448.00 ± 1.00	K	NIST Webbook
tc	614.84	K	Joback Method
tf	263.30	K	Joback Method
vc	0.330	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	164.07	J/mol×K	433.08	Joback Method
cpg	192.66	J/mol×K	584.55	Joback Method

cpg	187.23	J/molxK	554.25	Joback Method
cpg	181.65	J/molxK	523.96	Joback Method
cpg	175.92	J/molxK	493.67	Joback Method
cpg	170.05	J/molxK	463.37	Joback Method
cpg	197.92	J/molxK	614.84	Joback Method
dvisc	0.0003400	Paxs	433.08	Joback Method
dvisc	0.0004205	Paxs	404.78	Joback Method
dvisc	0.0005369	Paxs	376.49	Joback Method
dvisc	0.0007134	Paxs	348.19	Joback Method
dvisc	0.0009967	Paxs	319.89	Joback Method
dvisc	0.0014859	Paxs	291.60	Joback Method
dvisc	0.0024139	Paxs	263.30	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C629152&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C629152&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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