

# 1,1,1-Trifluorodecane

<b>Other names:</b>	Decane, 1,1,1-trifluoro
<b>Inchi:</b>	InChI=1S/C10H19F3/c1-2-3-4-5-6-7-8-9-10(11,12)13/h2-9H2,1H3
<b>InchiKey:</b>	IGGXVAUUTYVHBN-UHFFFAOYSA-N
<b>Formula:</b>	C10H19F3
<b>SMILES:</b>	CCCCCCCCC(F)(F)F
<b>Mol. weight [g/mol]:</b>	196.25
<b>CAS:</b>	26288-16-4

## Physical Properties

Property code	Value	Unit	Source
gf	-548.27	kJ/mol	Joback Method
hf	-846.81	kJ/mol	Joback Method
hfus	23.48	kJ/mol	Joback Method
hvap	34.11	kJ/mol	Joback Method
log10ws	-4.67		Crippen Method
logp	4.689		Crippen Method
mcvol	157.070	ml/mol	McGowan Method
pc	1900.26	kPa	Joback Method
rinpol	918.50		NIST Webbook
rinpol	922.00		NIST Webbook
tb	422.78	K	Joback Method
tc	571.35	K	Joback Method
tf	206.65	K	Joback Method
vc	0.638	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	341.37	J/molxK	422.78	Joback Method
cpg	355.77	J/molxK	447.54	Joback Method
cpg	369.59	J/molxK	472.30	Joback Method
cpg	382.84	J/molxK	497.07	Joback Method
cpg	395.54	J/molxK	521.83	Joback Method
cpg	407.70	J/molxK	546.59	Joback Method

cpg

419.36

J/mol×K

571.35

Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.29907e+01
Coeff. B	-3.38340e+03
Coeff. C	-6.06000e+01
Temperature range (K), min.	326.94
Temperature range (K), max.	501.19

## Sources

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=R172913&Units=SI>

The Yaws Handbook of Vapor Pressure:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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