

# Naphthalene, 1,1'-(1,10-decanediyl)bis[decahydro-

Other names:

1,10-Di-(1'-decahydronaphthyl)decane

1,10-Di-(decahydro-1-naphthyl)decane

1,10-bis(decahydro-1-naphthyl)decane

Naphthalene, 1,1'-(1,10-decanediyl)bis\*decahydro-

Inchi: InChI=1S/C30H54/c1(3-5-7-15-25-19-13-21-27-17-9-11-23-29(25)27)2-4-6-8-16-26-20-1

InchiKey: SHLQESUQERMUMR-UHFFFAOYSA-N

Formula: C30H54

SMILES: C(CCCCC1CCCC2CCCC12)CCCC1CCCC2CCCC12

Mol. weight [g/mol]: 414.75

CAS: 55268-64-9

## Physical Properties

Property code	Value	Unit	Source
gf	332.50	kJ/mol	Joback Method
hf	-461.29	kJ/mol	Joback Method
hfus	51.34	kJ/mol	Joback Method
hvap	82.78	kJ/mol	Joback Method
log10ws	-10.51		Crippen Method
logp	10.100		Crippen Method
mcvol	390.120	ml/mol	McGowan Method
pc	843.09	kPa	Joback Method
tb	937.58	K	Joback Method
tc	1156.99	K	Joback Method
tf	462.98	K	Joback Method
vc	1.478	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1601.86	J/molxK	1156.99	Joback Method
cpg	1583.68	J/molxK	1120.43	Joback Method
cpg	1564.14	J/molxK	1083.86	Joback Method
cpg	1543.13	J/molxK	1047.29	Joback Method
cpg	1520.51	J/molxK	1010.72	Joback Method

cpg	1496.15	J/mol×K	974.15	Joback Method
cpg	1469.93	J/mol×K	937.58	Joback Method
dvisc	0.0027444	Paxs	462.98	Joback Method
dvisc	0.0002260	Paxs	937.58	Joback Method
dvisc	0.0002829	Paxs	858.48	Joback Method
dvisc	0.0003706	Paxs	779.38	Joback Method
dvisc	0.0005159	Paxs	700.28	Joback Method
dvisc	0.0007815	Paxs	621.18	Joback Method
dvisc	0.0013362	Paxs	542.08	Joback Method
hvapt	119.70	kJ/mol	551.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.01695e+01
Coeff. B	-1.13167e+04
Coeff. C	-1.40010e+01
Temperature range (K), min.	583.19
Temperature range (K), max.	775.65

## Sources

**The Yaws Handbook of Vapor Pressure:**  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C55268649&Units=SI>

## Legend

**cpg:** Ideal gas heat capacity  
**dvisc:** Dynamic viscosity  
**gf:** Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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