

# Tetroquinone

<b>Other names:</b>	2,5-Cyclohexadiene-1,4-dione, 2,3,5,6-tetrahydroxy- p-Benzoquinone, 2,3,5,6-tetrahydroxy- HPEK-1 Kelox Terasin Tetrahydroxy-p-benzoquinone Tetrahydroxy-p-quinone Tetrahydroxy-1,4-benzoquinone Tetrahydroxyparabenzoquinone Tetrahydroxyquinone THQ 2,3,5,6-Tetrahydroxy-p-benzoquinone Tetrahydroxy-1,4-quinone 2,3,5,6-Tetrahydroxy-2,5-cyclohexadien-1,4-dione NSC-112931
<b>Inchi:</b>	InChI=1S/C6H4O6/c7-1-2(8)4(10)6(12)5(11)3(1)9/h7-8,11-12H
<b>InchiKey:</b>	DGQOCLATAPFASR-UHFFFAOYSA-N
<b>Formula:</b>	C6H4O6
<b>SMILES:</b>	O=C1C(O)=C(O)C(=O)C(O)=C1O
<b>Mol. weight [g/mol]:</b>	172.09
<b>CAS:</b>	319-89-1

## Physical Properties

Property code	Value	Unit	Source
gf	-739.26	kJ/mol	Joback Method
hf	-907.15	kJ/mol	Joback Method
hfus	18.32	kJ/mol	Joback Method
hvap	108.13	kJ/mol	Joback Method
log10ws	0.23		Crippen Method
logp	-0.207		Crippen Method
mcvol	102.560	ml/mol	McGowan Method
pc	8175.04	kPa	Joback Method
tb	883.50	K	Joback Method
tc	1088.58	K	Joback Method
tf	600.32	K	Joback Method
vc	0.367	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	292.83	J/mol×K	883.50	Joback Method
cpg	297.33	J/mol×K	917.68	Joback Method
cpg	301.20	J/mol×K	951.86	Joback Method
cpg	304.38	J/mol×K	986.04	Joback Method
cpg	306.83	J/mol×K	1020.22	Joback Method
cpg	308.51	J/mol×K	1054.40	Joback Method
cpg	309.38	J/mol×K	1088.58	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C319891&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C319891&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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