

# Perfluoromethyldiethylamine

<b>Inchi:</b>	InChI=1S/C5F13N/c6-1(7,8)3(12,13)19(5(16,17)18)4(14,15)2(9,10)11
<b>InchiKey:</b>	KMPITNDOTGKQSE-UHFFFAOYSA-N
<b>Formula:</b>	C5F13N
<b>SMILES:</b>	FC(F)(F)N(C(F)(F)C(F)(F)F)C(F)(F)C(F)(F)F
<b>Mol. weight [g/mol]:</b>	321.04
<b>CAS:</b>	758-48-5

## Physical Properties

Property code	Value	Unit	Source
gf	-2416.33	kJ/mol	Joback Method
hf	-2672.18	kJ/mol	Joback Method
hfus	14.70	kJ/mol	Joback Method
hvap	29.40 ± 0.40	kJ/mol	NIST Webbook
log10ws	-4.59		Crippen Method
logp	4.118		Crippen Method
mcvol	114.300	ml/mol	McGowan Method
pc	2085.03	kPa	Joback Method
sl	475.09	J/molxK	NIST Webbook
tb	300.60	K	Joback Method
tc	409.55	K	Joback Method
tf	150.10 ± 0.20	K	NIST Webbook
tt	149.64 ± 0.02	K	NIST Webbook
vc	0.512	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	245.41	J/molxK	300.60	Joback Method
cpg	257.16	J/molxK	318.76	Joback Method
cpg	268.28	J/molxK	336.92	Joback Method
cpg	278.80	J/molxK	355.07	Joback Method
cpg	288.75	J/molxK	373.23	Joback Method
cpg	298.13	J/molxK	391.39	Joback Method
cpg	306.96	J/molxK	409.55	Joback Method

cpl	337.46	J/mol×K	298.15	NIST Webbook
hfust	4.60	kJ/mol	150.10	NIST Webbook
hfust	7.16	kJ/mol	149.70	NIST Webbook
sfust	30.60	J/mol×K	150.10	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C758485&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C758485&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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