

# Acrylophenone, 3,3-diphenyl-

<b>Other names:</b>	Chalcone, «beta»-phenyl- 2-Propen-1-one, 1,3,3-triphenyl-
<b>Inchi:</b>	InChI=1S/C21H16O/c22-21(19-14-8-3-9-15-19)16-20(17-10-4-1-5-11-17)18-12-6-2-7-13
<b>InchiKey:</b>	PTKAYCBILQSQRK-UHFFFAOYSA-N
<b>Formula:</b>	C21H16O
<b>SMILES:</b>	O=C(C=C(c1ccccc1)c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	284.35
<b>CAS:</b>	849-01-4

## Physical Properties

Property code	Value	Unit	Source
chs	-10640.00	kJ/mol	NIST Webbook
gf	405.92	kJ/mol	Joback Method
hf	227.67	kJ/mol	Joback Method
hfs	81.38	kJ/mol	NIST Webbook
hfus	32.76	kJ/mol	Joback Method
hvap	75.95	kJ/mol	Joback Method
log10ws	-5.90		Crippen Method
logp	5.001		Crippen Method
mvol	232.740	ml/mol	McGowan Method
pc	2222.89	kPa	Joback Method
tb	817.83	K	Joback Method
tc	1087.12	K	Joback Method
tf	436.58	K	Joback Method
vc	0.875	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	648.57	J/molxK	817.83	Joback Method
cpg	664.57	J/molxK	862.71	Joback Method
cpg	679.12	J/molxK	907.59	Joback Method
cpg	692.41	J/molxK	952.48	Joback Method
cpg	704.64	J/molxK	997.36	Joback Method

cpg	716.00	J/mol×K	1042.24	Joback Method
cpg	726.67	J/mol×K	1087.12	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C849014&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C849014&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/77-525-4/Acrylophenone-3-3-diphenyl.pdf>

Generated by Cheméo on 2024-05-03 04:20:37.216280332 +0000 UTC m=+16999286.136857659.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.