

# Acetoin acetate

<b>Inchi:</b>	InChI=1S/C6H10O3/c1-4(7)5(2)9-6(3)8/h5H,1-3H3
<b>InchiKey:</b>	ZKPTYCJWRHHBOW-UHFFFAOYSA-N
<b>Formula:</b>	C6H10O3
<b>SMILES:</b>	CC(=O)OC(C)C(C)=O
<b>Mol. weight [g/mol]:</b>	130.14

## Physical Properties

Property code	Value	Unit	Source
gf	-365.64	kJ/mol	Joback Method
hf	-529.83	kJ/mol	Joback Method
hfus	12.16	kJ/mol	Joback Method
hvap	44.46	kJ/mol	Joback Method
log10ws	-0.59		Crippen Method
logp	0.527		Crippen Method
mcvol	104.410	ml/mol	McGowan Method
pc	3555.77	kPa	Joback Method
ripol	1389.00		NIST Webbook
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tb	466.40	K	Joback Method
tc	659.46	K	Joback Method
tf	264.47	K	Joback Method
vc	0.396	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	215.20	J/mol×K	466.40	Joback Method
cpg	224.83	J/mol×K	498.58	Joback Method
cpg	234.10	J/mol×K	530.75	Joback Method
cpg	243.00	J/mol×K	562.93	Joback Method
cpg	251.53	J/mol×K	595.11	Joback Method
cpg	259.68	J/mol×K	627.28	Joback Method
cpg	267.46	J/mol×K	659.46	Joback Method
dvisc	0.0036503	Paxs	264.47	Joback Method

dvisc	0.0019016	Paxs	298.12	Joback Method
dvisc	0.0011308	Paxs	331.78	Joback Method
dvisc	0.0007400	Paxs	365.44	Joback Method
dvisc	0.0005201	Paxs	399.09	Joback Method
dvisc	0.0003862	Paxs	432.75	Joback Method
dvisc	0.0002994	Paxs	466.40	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R610376&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R610376&amp;Units=SI</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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