

# p-Menth-3-en-2-ol

Inchi:	InChI=1S/C10H18O/c1-7(2)9-5-4-8(3)10(11)6-9/h6-8,10-11H,4-5H2,1-3H3
InchiKey:	WEJSIMPWEOUMJB-UHFFFAOYSA-N
Formula:	C10H18O
SMILES:	CC(C)C1=CC(O)C(C)CC1
Mol. weight [g/mol]:	154.25

## Physical Properties

Property code	Value	Unit	Source
gf	-68.87	kJ/mol	Joback Method
hf	-326.95	kJ/mol	Joback Method
hfus	15.96	kJ/mol	Joback Method
hvap	55.22	kJ/mol	Joback Method
log10ws	-2.65		Crippen Method
logp	2.360		Crippen Method
mcvol	142.470	ml/mol	McGowan Method
pc	2814.34	kPa	Joback Method
rinpol	1205.00		NIST Webbook
rinpol	1186.00		NIST Webbook
rinpol	1186.00		NIST Webbook
rinpol	1202.00		NIST Webbook
rinpol	1202.00		NIST Webbook
ripol	1888.00		NIST Webbook
tb	538.96	K	Joback Method
tc	731.46	K	Joback Method
tf	264.70	K	Joback Method
vc	0.526	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	351.74	J/molxK	538.96	Joback Method
cpg	423.14	J/molxK	699.37	Joback Method
cpg	410.31	J/molxK	667.29	Joback Method
cpg	396.76	J/molxK	635.21	Joback Method

cpg	382.50	J/mol×K	603.13	Joback Method
cpg	367.50	J/mol×K	571.04	Joback Method
cpg	435.28	J/mol×K	731.46	Joback Method
dvisc	0.0001244	Paxs	538.96	Joback Method
dvisc	0.0001992	Paxs	493.25	Joback Method
dvisc	0.0003512	Paxs	447.54	Joback Method
dvisc	0.0007045	Paxs	401.83	Joback Method
dvisc	0.0016898	Paxs	356.12	Joback Method
dvisc	0.0052446	Paxs	310.41	Joback Method
dvisc	0.0240696	Paxs	264.70	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R325187&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R325187&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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