

# Spiro[5.6]dodecane

<b>Other names:</b>	Spiro(5,6)dodecane
<b>Inchi:</b>	InChI=1S/C12H22/c1-2-5-9-12(8-4-1)10-6-3-7-11-12/h1-11H2
<b>InchiKey:</b>	ZDNANLAHKKNCTC-UHFFFAOYSA-N
<b>Formula:</b>	C12H22
<b>SMILES:</b>	C1CCCC2(CC1)CCCC2
<b>Mol. weight [g/mol]:</b>	166.30
<b>CAS:</b>	181-15-7

## Physical Properties

Property code	Value	Unit	Source
gf	101.28	kJ/mol	Joback Method
hf	-146.79	kJ/mol	Joback Method
hfus	3.14	kJ/mol	Joback Method
hvap	42.32	kJ/mol	Joback Method
log10ws	-4.39		Crippen Method
logp	4.291		Crippen Method
mcvol	158.220	ml/mol	McGowan Method
pc	2823.32	kPa	Joback Method
tb	517.97	K	Joback Method
tc	760.42	K	Joback Method
tf	267.90	K	Joback Method
vc	0.573	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	385.02	J/mol×K	517.97	Joback Method
cpg	410.75	J/mol×K	558.38	Joback Method
cpg	434.58	J/mol×K	598.79	Joback Method
cpg	456.72	J/mol×K	639.20	Joback Method
cpg	477.33	J/mol×K	679.60	Joback Method
cpg	496.62	J/mol×K	720.01	Joback Method
cpg	514.77	J/mol×K	760.42	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.81411e+01
Coeff. B	-1.18250e+04
Temperature range (K), min.	424.54
Temperature range (K), max.	517.97

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C181157&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C181157&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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