

# t-Butylhydroquinone

<b>Other names:</b>	1,4-Benzenediol, 2-(1,1-dimethylethyl)-Hydroquinone, tert-butyl-tert-Butyl-1,4-Benzenediol tert-Butylhydroquinone Mono-tert-butylhydroquinone MTBHQ TBHQ Mono-tertiarybutylhydroquinone 2-tert-Butyl-1,4-benzenediol 2-tert-Butylhydroquinone Sustane Tenox TBHQ 2-t-Butyl-1,4-benzenediol 2-(1,1-Dimethylethyl)-1,4-benzenediol Eastman MTBHQ 2-t-Butylhydroquinone Banox 20BA NSC 4972 2-tert-butyl-1,4-dihydroxybenzene
<b>Inchi:</b>	InChI=1S/C10H14O2/c1-10(2,3)8-6-7(11)4-5-9(8)12/h4-6,11-12H,1-3H3
<b>InchiKey:</b>	BGNXCDMCOKJUMV-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O2
<b>SMILES:</b>	CC(C)(C)c1cc(O)ccc1O
<b>Mol. weight [g/mol]:</b>	166.22
<b>CAS:</b>	1948-33-0

## Physical Properties

Property code	Value	Unit	Source
gf	-160.67	kJ/mol	Joback Method
hf	-376.57	kJ/mol	Joback Method
hfus	19.85	kJ/mol	Joback Method
hsub	104.40 ± 1.30	kJ/mol	NIST Webbook
hvap	64.86	kJ/mol	Joback Method
log10ws	-1.88		Crippen Method
logp	2.395		Crippen Method
mcvol	139.740	ml/mol	McGowan Method
pc	4271.86	kPa	Joback Method

rmpol	1566.00		NIST Webbook
tb	612.89	K	Joback Method
tc	856.37	K	Joback Method
tf	454.74	K	Joback Method
vc	0.408	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	364.71	J/mol×K	612.89	Joback Method
cpg	377.38	J/mol×K	653.47	Joback Method
cpg	389.01	J/mol×K	694.05	Joback Method
cpg	399.81	J/mol×K	734.63	Joback Method
cpg	409.96	J/mol×K	775.21	Joback Method
cpg	419.64	J/mol×K	815.79	Joback Method
cpg	429.06	J/mol×K	856.37	Joback Method
dvisc	0.0000897	Paxs	481.10	Joback Method
dvisc	0.0001976	Paxs	454.74	Joback Method
dvisc	0.0000442	Paxs	507.46	Joback Method
dvisc	0.0000234	Paxs	533.82	Joback Method
dvisc	0.0000131	Paxs	560.17	Joback Method
dvisc	0.0000078	Paxs	586.53	Joback Method
dvisc	0.0000048	Paxs	612.89	Joback Method
hfust	27.74	kJ/mol	350.90	NIST Webbook
hsubt	101.20 ± 1.30	kJ/mol	350.50	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1948330&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1948330&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/75-920-7/t-Butylhydroquinone.pdf>

Generated by Cheméo on 2024-04-23 20:02:33.777901138 +0000 UTC m=+16191802.698478449.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.