

# 9-methyldecane-3-one

<b>Inchi:</b>	InChI=1S/C11H22O/c1-4-11(12)9-7-5-6-8-10(2)3/h10H,4-9H2,1-3H3
<b>InchiKey:</b>	UMUMYGFFTAWUQZ-UHFFFAOYSA-N
<b>Formula:</b>	C11H22O
<b>SMILES:</b>	CCC(=O)CCCCC(C)C
<b>Mol. weight [g/mol]:</b>	170.29

## Physical Properties

Property code	Value	Unit	Source
gf	-89.62	kJ/mol	Joback Method
hf	-388.23	kJ/mol	Joback Method
hfus	22.32	kJ/mol	Joback Method
hvap	46.44	kJ/mol	Joback Method
log10ws	-3.47		Crippen Method
logp	3.572		Crippen Method
mcvol	167.420	ml/mol	McGowan Method
pc	2060.49	kPa	Joback Method
rinpol	1260.00		NIST Webbook
rinpol	1260.00		NIST Webbook
tb	504.51	K	Joback Method
tc	679.38	K	Joback Method
tf	248.66	K	Joback Method
vc	0.651	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	388.41	J/mol×K	504.51	Joback Method
cpg	460.10	J/mol×K	650.23	Joback Method
cpg	446.99	J/mol×K	621.09	Joback Method
cpg	433.29	J/mol×K	591.94	Joback Method
cpg	418.97	J/mol×K	562.80	Joback Method
cpg	404.02	J/mol×K	533.65	Joback Method
cpg	472.61	J/mol×K	679.38	Joback Method
dvisc	0.0002371	Paxs	504.51	Joback Method

dvisc	0.0003195	Paxs	461.87	Joback Method
dvisc	0.0004574	Paxs	419.23	Joback Method
dvisc	0.0007105	Paxs	376.59	Joback Method
dvisc	0.0012349	Paxs	333.94	Joback Method
dvisc	0.0025234	Paxs	291.30	Joback Method
dvisc	0.0065882	Paxs	248.66	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R405451&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R405451&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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