

# Phenol, 4-(1-phenylethyl)-

<b>Other names:</b>	Phenol, p-(«alpha»-methylbenzyl)- 1-(4-Hydroxyphenyl)-1-phenylethane 1-Phenyl-1-(4-hydroxyphenyl)ethane 4-(«alpha»-Methylbenzyl)phenol 4-(Phenylethylidene)phenol 4-(1-Phenylethyl)phenol 4-(1-Fenylethyl)fenol p-(«alpha»-Methylbenzyl)phenol Styrolfenol NSC 1859 NSC 60586 p-(1-phenylethyl)phenol
<b>Inchi:</b>	InChI=1S/C14H14O/c1-11(12-5-3-2-4-6-12)13-7-9-14(15)10-8-13/h2-11,15H,1H3
<b>InchiKey:</b>	XHASMJXNUHCHBL-UHFFFAOYSA-N
<b>Formula:</b>	C14H14O
<b>SMILES:</b>	CC(c1ccccc1)c1ccc(O)cc1
<b>Mol. weight [g/mol]:</b>	198.26
<b>CAS:</b>	1988-89-2

## Physical Properties

Property code	Value	Unit	Source
gf	134.76	kJ/mol	Joback Method
hf	-41.82	kJ/mol	Joback Method
hfus	22.36	kJ/mol	Joback Method
hvap	63.94	kJ/mol	Joback Method
log10ws	-3.50		Crippen Method
logp	3.544		Crippen Method
mcvol	166.470	ml/mol	McGowan Method
pc	3291.59	kPa	Joback Method
tb	653.26	K	Joback Method
tc	907.43	K	Joback Method
tf	397.10	K	Joback Method
vc	0.564	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	427.10	J/molxK	653.26	Joback Method
cpg	442.76	J/molxK	695.62	Joback Method
cpg	457.16	J/molxK	737.98	Joback Method
cpg	470.46	J/molxK	780.35	Joback Method
cpg	482.79	J/molxK	822.71	Joback Method
cpg	494.33	J/molxK	865.07	Joback Method
cpg	505.21	J/molxK	907.43	Joback Method
dvisc	0.0004594	Paxs	439.79	Joback Method
dvisc	0.0012972	Paxs	397.10	Joback Method
dvisc	0.0001955	Paxs	482.49	Joback Method
dvisc	0.0000956	Paxs	525.18	Joback Method
dvisc	0.0000521	Paxs	567.87	Joback Method
dvisc	0.0000309	Paxs	610.57	Joback Method
dvisc	0.0000196	Paxs	653.26	Joback Method
hvapt	90.80	kJ/mol	482.00	NIST Webbook
hvapt	75.40	kJ/mol	485.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1988892&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1988892&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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