

# 3,5,3',5'-Tetramethylbiphenyl

<b>Other names:</b>	1,1'-Biphenyl, 3,3',5,5'-tetramethyl-Biphenyl, 3,3',5,5'-tetramethyl-3,3',5,5'-Tetramethylbiphenyl, 2,2'5.5'-Tetramethylbiphenyl
<b>Inchi:</b>	InChI=1S/C16H18/c1-11-5-12(2)8-15(7-11)16-9-13(3)6-14(4)10-16/h5-10H,1-4H3
<b>InchiKey:</b>	CMZYGFLQKQMKF-UHFFFAOYSA-N
<b>Formula:</b>	C16H18
<b>SMILES:</b>	<chem>Cc1cc(C)cc(-c2cc(C)cc(C)c2)c1</chem>
<b>Mol. weight [g/mol]:</b>	210.31
<b>CAS:</b>	25570-02-9

## Physical Properties

Property code	Value	Unit	Source
gf	270.14	kJ/mol	Joback Method
hf	53.61	kJ/mol	Joback Method
hfus	23.72	kJ/mol	Joback Method
hvap	58.41	kJ/mol	Joback Method
log10ws	-5.98		Crippen Method
logp	4.587		Crippen Method
mcvol	188.780	ml/mol	McGowan Method
pc	2159.31	kPa	Joback Method
tb	638.76	K	Joback Method
tc	872.54	K	Joback Method
tf	322.00 ± 2.00	K	NIST Webbook
vc	0.716	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	470.87	J/mol×K	638.76	Joback Method
cpg	488.57	J/mol×K	677.72	Joback Method
cpg	505.13	J/mol×K	716.69	Joback Method
cpg	520.59	J/mol×K	755.65	Joback Method
cpg	535.01	J/mol×K	794.62	Joback Method

cpg	548.43	J/mol×K	833.58	Joback Method
cpg	560.89	J/mol×K	872.54	Joback Method
dvisc	0.0009410	Paxs	373.00	Joback Method
dvisc	0.0005836	Paxs	417.29	Joback Method
dvisc	0.0003967	Paxs	461.59	Joback Method
dvisc	0.0002885	Paxs	505.88	Joback Method
dvisc	0.0002209	Paxs	550.17	Joback Method
dvisc	0.0001759	Paxs	594.47	Joback Method
dvisc	0.0001447	Paxs	638.76	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C25570029&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C25570029&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/74-030-6/3-5-3-5-Tetramethylbiphenyl.pdf>

Generated by Cheméo on 2024-05-09 00:08:15.66694394 +0000 UTC m=+17502544.587521256.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.