

3,6-Dimethyl-6-formyl-5,6-dihydropyran

Inchi:	InChI=1S/C9H14O/c1-4-9(3)6-5-8(2)7-10-9/h4-5H,1,6-7H2,2-3H3
InchiKey:	VBWAGXIOADKMRE-UHFFFAOYSA-N
Formula:	C9H14O
SMILES:	C=CC1(C)CC=C(C)CO1
Mol. weight [g/mol]:	138.21

Physical Properties

Property code	Value	Unit	Source
gf	65.91	kJ/mol	Joback Method
hf	-119.79	kJ/mol	Joback Method
hfus	12.13	kJ/mol	Joback Method
hvap	39.70	kJ/mol	Joback Method
log10ws	-2.39		Crippen Method
logp	2.298		Crippen Method
mcvol	124.080	ml/mol	McGowan Method
pc	3163.27	kPa	Joback Method
ripol	1709.00		NIST Webbook
ripol	1709.00		NIST Webbook
tb	452.88	K	Joback Method
tc	668.72	K	Joback Method
tf	260.56	K	Joback Method
vc	0.459	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	258.73	J/mol×K	452.88	Joback Method
cpg	274.74	J/mol×K	488.85	Joback Method
cpg	289.62	J/mol×K	524.83	Joback Method
cpg	303.49	J/mol×K	560.80	Joback Method
cpg	316.46	J/mol×K	596.78	Joback Method
cpg	328.62	J/mol×K	632.75	Joback Method
cpg	340.10	J/mol×K	668.72	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R560771&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
riol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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