

# 1-Heptyn-4-ol

<b>Inchi:</b>	InChI=1S/C7H12O/c1-3-5-7(8)6-4-2/h1,7-8H,4-6H2,2H3
<b>InchiKey:</b>	LYXYBSYXGARUEA-UHFFFAOYSA-N
<b>Formula:</b>	C7H12O
<b>SMILES:</b>	C#CCC(O)CCC
<b>Mol. weight [g/mol]:</b>	112.17
<b>CAS:</b>	22127-83-9

## Physical Properties

Property code	Value	Unit	Source
gf	91.87	kJ/mol	Joback Method
hf	-53.42	kJ/mol	Joback Method
hfus	17.43	kJ/mol	Joback Method
hvap	47.33	kJ/mol	Joback Method
log10ws	-1.92		Crippen Method
logp	1.171		Crippen Method
mcvol	106.760	ml/mol	McGowan Method
pc	3708.97	kPa	Joback Method
tb	441.42	K	Joback Method
tc	615.96	K	Joback Method
tf	261.44	K	Joback Method
vc	0.403	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	218.57	J/molxK	441.42	Joback Method
cpg	228.02	J/molxK	470.51	Joback Method
cpg	237.04	J/molxK	499.60	Joback Method
cpg	245.67	J/molxK	528.69	Joback Method
cpg	253.91	J/molxK	557.78	Joback Method
cpg	261.78	J/molxK	586.87	Joback Method
cpg	269.29	J/molxK	615.96	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.64196e+01
Coeff. B	-4.55664e+03
Coeff. C	-6.71610e+01
Temperature range (K), min.	349.15
Temperature range (K), max.	477.15

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="https://webbook.nist.gov/cgi/cbook.cgi?ID=C22127839&amp;Units=SI">https://webbook.nist.gov/cgi/cbook.cgi?ID=C22127839&amp;Units=SI</a>
The Yaws Handbook of Vapor Pressure:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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