

# 4,4'-Dinitrobibenzyl

**Other names:**

4,4'-Dinitrodibenzyl  
Benzene, 1,1'-(1,2-ethanediyl)bis[4-nitro-  
p,p'-Dinitrobibenzyl  
p,p'-Dinitrodibenzyl  
Bibenzyl, 4,4'-dinitro-  
1,2-Bis(p-nitrophenyl)ethane  
1,2-Bis(4-nitrophenyl)ethane  
1,2-Ethanediylbis(4-nitrobenzene)  
1,1'-(1,2-Ethanediyl)bis(4-nitrobenzene)  
NSC 60022  
NSC 608

**Inchi:**

InChI=1S/C14H12N2O4/c17-15(18)13-7-3-11(4-8-13)1-2-12-5-9-14(10-6-12)16(19)20/h3

**InchiKey:**

BSNKHEKTRWNNAC-UHFFFAOYSA-N

**Formula:**

C<sub>14</sub>H<sub>12</sub>N<sub>2</sub>O<sub>4</sub>

**SMILES:**

O=[N+](O)c1ccc(Cc2ccc([N+](=O)[O-])cc2)cc1

**Mol. weight [g/mol]:**

272.26

**CAS:**

736-30-1

## Physical Properties

Property code	Value	Unit	Source
gf	343.66	kJ/mol	Joback Method
hf	96.31	kJ/mol	Joback Method
hfus	42.04	kJ/mol	Joback Method
hvap	85.82	kJ/mol	Joback Method
ie	9.50 ± 0.10	eV	NIST Webbook
ie	9.77 ± 0.05	eV	NIST Webbook
log10ws	-5.19		Crippen Method
logp	3.288		Crippen Method
mcvol	195.440	ml/mol	McGowan Method
pc	2752.67	kPa	Joback Method
rinpol	419.60		NIST Webbook
tb	886.72	K	Joback Method
tc	1163.96	K	Joback Method
tf	612.64	K	Joback Method
vc	0.767	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	565.16	J/molxK	886.72	Joback Method
cpg	576.51	J/molxK	932.93	Joback Method
cpg	586.69	J/molxK	979.13	Joback Method
cpg	595.81	J/molxK	1025.34	Joback Method
cpg	604.00	J/molxK	1071.54	Joback Method
cpg	611.37	J/molxK	1117.75	Joback Method
cpg	618.04	J/molxK	1163.96	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C736301&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C736301&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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