

# Benzeneacetonitrile, «alpha»-phenyl-

<b>Other names:</b>	Acetonitrile, diphenyl- «alpha»-Phenylbenzylcyanide «alpha»-Phenylphenylacetonitrile Benzhydrylcyanide Dipan Diphenatril Diphenyl-«alpha»-cyanomethane Diphenylacetonitrile Diphenylmethylcyanide USAF KF-13 Benzyhydrylcyanide «alpha»-Cyanodiphenylmethane Difenylacetonitril alpha-Phenylbenzyl cyanide alpha-Phenylphenylacetonitrile Diphenyl-alpha-cyanomethane «alpha»-Phenyl-benzeneacetonitrile NSC 130268
<b>Inchi:</b>	InChI=1S/C14H11N/c15-11-14(12-7-3-1-4-8-12)13-9-5-2-6-10-13/h1-10,14H
<b>InchiKey:</b>	NEBPTMCRLHKPOB-UHFFFAOYSA-N
<b>Formula:</b>	C14H11N
<b>SMILES:</b>	N#CC(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	193.24
<b>CAS:</b>	86-29-3

## Physical Properties

Property code	Value	Unit	Source
gf	422.56	kJ/mol	Joback Method
hf	300.37	kJ/mol	Joback Method
hfus	18.08	kJ/mol	Joback Method
hvap	61.40	kJ/mol	Joback Method
log10ws	-3.82		Crippen Method
logp	3.342		Crippen Method
mcvol	161.980	ml/mol	McGowan Method
pc	2721.17	kPa	Joback Method
tb	674.72	K	Joback Method
tc	932.92	K	Joback Method

tf	350.37	K	Joback Method
vc	0.624	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	400.81	J/mol×K	674.72	Joback Method
cpg	415.07	J/mol×K	717.75	Joback Method
cpg	428.07	J/mol×K	760.79	Joback Method
cpg	439.89	J/mol×K	803.82	Joback Method
cpg	450.64	J/mol×K	846.86	Joback Method
cpg	460.40	J/mol×K	889.89	Joback Method
cpg	469.28	J/mol×K	932.92	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	454.20	K	1.60	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C86293&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C86293&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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