

Ethyne, fluoro-

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|-----------------------------|---|
| Other names: | Fluoroacetylene Acetylene, fluoro- Monofluoroacetylene HC«equiv»CF Fluoroethyne C2HF |
| Inchi: | InChI=1S/C2HF/c1-2-3/h1H |
| InchiKey: | IAWCIZWLKMTPLL-UHFFFAOYSA-N |
| Formula: | C2HF |
| SMILES: | C#CF |
| Mol. weight [g/mol]: | 44.03 |
| CAS: | 2713-09-9 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|--------------|----------------------|----------------|
| affp | 686.00 | kJ/mol | NIST Webbook |
| basg | 661.30 | kJ/mol | NIST Webbook |
| gf | -5.78 | kJ/mol | Joback Method |
| hf | 11.18 | kJ/mol | Joback Method |
| hfus | 6.99 | kJ/mol | Joback Method |
| hvap | 19.09 | kJ/mol | Joback Method |
| ie | 11.50 ± 0.10 | eV | NIST Webbook |
| ie | 11.26 | eV | NIST Webbook |
| ie | 11.26 | eV | NIST Webbook |
| ie | 11.26 | eV | NIST Webbook |
| log10ws | -0.80 | | Crippen Method |
| logp | 0.547 | | Crippen Method |
| mcvol | 32.210 | ml/mol | McGowan Method |
| pc | 5853.95 | kPa | Joback Method |
| tb | 234.55 | K | Joback Method |
| tc | 391.57 | K | Joback Method |
| tf | 159.86 | K | Joback Method |
| vc | 0.128 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-------|---------|-----------------|---------------|
| cpg | 42.26 | J/mol×K | 234.55 | Joback Method |
| cpg | 44.09 | J/mol×K | 260.72 | Joback Method |
| cpg | 45.84 | J/mol×K | 286.89 | Joback Method |
| cpg | 47.49 | J/mol×K | 313.06 | Joback Method |
| cpg | 49.06 | J/mol×K | 339.23 | Joback Method |
| cpg | 50.55 | J/mol×K | 365.40 | Joback Method |
| cpg | 51.95 | J/mol×K | 391.57 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C2713099&Units=SI |

Legend

| | |
|-----------------|---|
| affp: | Proton affinity |
| basg: | Gas basicity |
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| ie: | Ionization energy |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |

vc: Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/69-949-3/Ethyne-fluoro.pdf>

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