

# 2,4,6-Tripicryltriazine

**Inchi:** InChI=1S/C21H6N12O18/c34-25(35)7-1-10(28(40)41)16(11(2-7)29(42)43)19-22-20(17-1  
**InchiKey:** FBPVHYPYBVZWJZ-UHFFFAOYSA-N  
**Formula:** C21H6N12O18  
**SMILES:** O=[N+]([O-])c1cc([N+](=O)[O-])c(-c2nc(-c3c([N+](=O)[O-])cc([N+](=O)[O-])cc3[N+](=O)[O-])  
**Mol. weight [g/mol]:** 714.34  
**CAS:** 49753-54-0

## Physical Properties

Property code	Value	Unit	Source
chs	-9417.70 ± 7.00	kJ/mol	NIST Webbook
chs	-9439.10 ± 7.10	kJ/mol	NIST Webbook
hfs	296.50 ± 7.00	kJ/mol	NIST Webbook
hfs	318.00 ± 17.00	kJ/mol	NIST Webbook
log10ws	-14.43		Crippen Method
logp	4.046		Crippen Method
mcpol	398.430	ml/mol	McGowan Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hsubt	167.90	kJ/mol	515.00	NIST Webbook

## Sources

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)  
**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C49753540&Units=SI>  
**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

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