

2,4',6-Trichloro-1,1'-biphenyl

Other names:	1,1'-Biphenyl, 2,4',6-trichloro 1,3-dichloro-2-(4-chlorophenyl)benzene 2,4,6-PCB PCB 32
Inchi:	InChI=1S/C12H7Cl3/c13-9-6-4-8(5-7-9)12-10(14)2-1-3-11(12)15/h1-7H
InchiKey:	IHIDFKLAWYPTKB-UHFFFAOYSA-N
Formula:	C12H7Cl3
SMILES:	Clc1ccc(-c2c(Cl)cccc2Cl)cc1
Mol. weight [g/mol]:	257.54
CAS:	38444-77-8

Physical Properties

Property code	Value	Unit	Source
gf	210.30	kJ/mol	Joback Method
hf	100.42	kJ/mol	Joback Method
hfus	26.34	kJ/mol	Joback Method
hvap	62.00	kJ/mol	Joback Method
log10ws	-6.14		Aqueous Solubility Prediction Method
log10ws	-6.14		Estimated Solubility Method
logp	5.314		Crippen Method
mcvol	169.140	ml/mol	McGowan Method
pc	2878.12	kPa	Joback Method
rinpol	1767.00		NIST Webbook
rinpol	1777.00		NIST Webbook
rinpol	1837.00		NIST Webbook
rinpol	1841.00		NIST Webbook
rinpol	1770.00		NIST Webbook
rinpol	1772.00		NIST Webbook
rinpol	1836.00		NIST Webbook
rinpol	1836.00		NIST Webbook
rinpol	1836.00		NIST Webbook
rinpol	1797.00		NIST Webbook
tb	654.55	K	Joback Method
tc	917.95	K	Joback Method
tf	405.16	K	Joback Method
vc	0.638	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	349.20	J/molxK	654.55	Joback Method
cpg	360.92	J/molxK	698.45	Joback Method
cpg	371.61	J/molxK	742.35	Joback Method
cpg	381.33	J/molxK	786.25	Joback Method
cpg	390.14	J/molxK	830.15	Joback Method
cpg	398.13	J/molxK	874.05	Joback Method
cpg	405.36	J/molxK	917.95	Joback Method
dvisc	0.0010725	Paxs	405.16	Joback Method
dvisc	0.0006985	Paxs	446.73	Joback Method
dvisc	0.0004893	Paxs	488.29	Joback Method
dvisc	0.0003625	Paxs	529.86	Joback Method
dvisc	0.0002805	Paxs	571.42	Joback Method
dvisc	0.0002248	Paxs	612.99	Joback Method
dvisc	0.0001852	Paxs	654.55	Joback Method

Sources

Estimated Solubility Method:	http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C38444778&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
Aqueous Solubility Prediction Method:	http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
m_{cvol}:	McGowan's characteristic volume
p_c:	Critical Pressure
r_{inpol}:	Non-polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/68-346-3/2-4-6-Trichloro-1-1-biphenyl.pdf>

Generated by Cheméo on 2024-04-17 03:05:16.974930488 +0000 UTC m=+15612365.895507804.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.