

# 2-Ketobutan-3-yl butanoate

<b>Other names:</b>	Acetoin butanoate 3-(butanoyloxy)-2-butanone
<b>Inchi:</b>	InChI=1S/C8H14O3/c1-4-5-8(10)11-7(3)6(2)9/h7H,4-5H2,1-3H3
<b>InchiKey:</b>	LJDWJXUIGKSETE-UHFFFAOYSA-N
<b>Formula:</b>	C8H14O3
<b>SMILES:</b>	CCCC(=O)OC(C)C(C)=O
<b>Mol. weight [g/mol]:</b>	158.19

## Physical Properties

Property code	Value	Unit	Source
gf	-348.80	kJ/mol	Joback Method
hf	-571.11	kJ/mol	Joback Method
hfus	17.34	kJ/mol	Joback Method
hvap	48.92	kJ/mol	Joback Method
log10ws	-1.43		Crippen Method
logp	1.307		Crippen Method
mcvol	132.590	ml/mol	McGowan Method
pc	2862.74	kPa	Joback Method
rinpol	1036.00		NIST Webbook
rinpol	1038.00		NIST Webbook
rinpol	1044.00		NIST Webbook
rinpol	1038.00		NIST Webbook
tb	512.16	K	Joback Method
tc	700.78	K	Joback Method
tf	287.01	K	Joback Method
vc	0.507	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	298.94	J/molxK	512.16	Joback Method
cpg	310.84	J/molxK	543.60	Joback Method
cpg	322.24	J/molxK	575.03	Joback Method
cpg	333.16	J/molxK	606.47	Joback Method

cpg	343.58	J/mol×K	637.91	Joback Method
cpg	353.52	J/mol×K	669.35	Joback Method
cpg	362.98	J/mol×K	700.78	Joback Method
dvisc	0.0036492	Paxs	287.01	Joback Method
dvisc	0.0018279	Paxs	324.53	Joback Method
dvisc	0.0010567	Paxs	362.06	Joback Method
dvisc	0.0006771	Paxs	399.58	Joback Method
dvisc	0.0004683	Paxs	437.11	Joback Method
dvisc	0.0003434	Paxs	474.63	Joback Method
dvisc	0.0002635	Paxs	512.16	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R66269&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R66269&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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