

# Cyclohexanol, 2-amino-, cis-

<b>Other names:</b>	cis-2-Aminocyclohexanol 2-Aminocyclohexanol, cis
<b>Inchi:</b>	InChI=1S/C6H13NO/c7-5-3-1-2-4-6(5)8/h5-6,8H,1-4,7H2/t5-,6+/m1/s1
<b>InchiKey:</b>	PQMCFTMVQORYJC-RITPCOANSA-N
<b>Formula:</b>	C6H13NO
<b>SMILES:</b>	NC1CCCCC1O
<b>Mol. weight [g/mol]:</b>	115.17
<b>CAS:</b>	931-15-7

## Physical Properties

Property code	Value	Unit	Source
gf	-53.99	kJ/mol	Joback Method
hf	-251.63	kJ/mol	Joback Method
hfus	13.49	kJ/mol	Joback Method
hvap	56.39	kJ/mol	Joback Method
ie	9.59	eV	NIST Webbook
log10ws	-1.15		Crippen Method
logp	0.249		Crippen Method
mcvol	100.390	ml/mol	McGowan Method
pc	4608.87	kPa	Joback Method
tb	516.27	K	Joback Method
tc	722.46	K	Joback Method
tf	380.50 ± 0.50	K	NIST Webbook
vc	0.351	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	246.09	J/molxK	516.27	Joback Method
cpg	259.48	J/molxK	550.64	Joback Method
cpg	272.17	J/molxK	585.00	Joback Method
cpg	284.16	J/molxK	619.37	Joback Method
cpg	295.48	J/molxK	653.73	Joback Method
cpg	306.14	J/molxK	688.10	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	382.00 ± 1.00	K	2.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C931157&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C931157&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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