

# 1H-Tetrazole, 1-phenyl-

<b>Other names:</b>	1-Phenyl-1H-tetrazole 1-Phenyltetrazole
<b>Inchi:</b>	InChI=1S/C7H6N4/c1-2-4-7(5-3-1)11-6-8-9-10-11/h1-6H
<b>InchiKey:</b>	IYPXPGSELZFFMI-UHFFFAOYSA-N
<b>Formula:</b>	C7H6N4
<b>SMILES:</b>	c1ccc(-n2cnnc2)cc1
<b>Mol. weight [g/mol]:</b>	146.15
<b>CAS:</b>	5378-52-9

## Physical Properties

Property code	Value	Unit	Source
chs	-3973.90 ± 1.50	kJ/mol	NIST Webbook
chs	-3956.00 ± 0.40	kJ/mol	NIST Webbook
hf	448.00 ± 3.00	kJ/mol	NIST Webbook
hf	465.80	kJ/mol	NIST Webbook
hfs	344.00 ± 0.40	kJ/mol	NIST Webbook
hfs	361.80	kJ/mol	NIST Webbook
hsub	104.00 ± 3.00	kJ/mol	NIST Webbook
hsub	104.00	kJ/mol	NIST Webbook
log10ws	-2.42		Crippen Method
logp	0.662		Crippen Method
mvol	106.190	ml/mol	McGowan Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5378529&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5378529&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

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