

# Cyclohexanethiol

<b>Other names:</b>	CYCLOHEXYL MERCAPTAN CYCLOHEXYL THIOL Cyklohexanthiol Cyklohexylmerkaptan
<b>Inchi:</b>	InChI=1S/C6H12S/c7-6-4-2-1-3-5-6/h6-7H,1-5H2
<b>InchiKey:</b>	CMKBCTPCXZNQKX-UHFFFAOYSA-N
<b>Formula:</b>	C6H12S
<b>SMILES:</b>	SC1CCCCC1
<b>Mol. weight [g/mol]:</b>	116.22
<b>CAS:</b>	1569-69-3

## Physical Properties

Property code	Value	Unit	Source
af	0.2520		KDB
chl	-4537.60 ± 3.60	kJ/mol	NIST Webbook
gf	53.48	kJ/mol	Joback Method
hf	-95.73 ± 0.79	kJ/mol	NIST Webbook
hf	-95.69	kJ/mol	NIST Webbook
hfl	-140.40 ± 0.79	kJ/mol	NIST Webbook
hfus	7.17	kJ/mol	Joback Method
hvap	44.90	kJ/mol	NIST Webbook
hvap	44.70	kJ/mol	NIST Webbook
hvap	44.60 ± 0.20	kJ/mol	NIST Webbook
hvap	44.60 ± 0.10	kJ/mol	NIST Webbook
hvap	44.57	kJ/mol	NIST Webbook
log10ws	-2.41		Crippen Method
logp	2.249		Crippen Method
mvol	100.890	ml/mol	McGowan Method
pc	3850.00	kPa	KDB
rinpol	983.00		NIST Webbook
rinpol	983.00		NIST Webbook
rinpol	983.00		NIST Webbook
sl	255.57	J/molxK	NIST Webbook
tb	431.98	K	KDB
tb	432.00	K	NIST Webbook
tb	431.95 ± 0.20	K	NIST Webbook
tb	427.00 ± 4.00	K	NIST Webbook

tb	428.00 ± 3.00	K	NIST Webbook
tb	432.20	K	NIST Webbook
tc	664.00	K	NIST Webbook
tc	664.20	K	KDB
tf	201.22	K	Joback Method
tt	189.63 ± 0.01	K	NIST Webbook
vc	0.358	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	184.05	J/mol×K	419.09	Joback Method
cpg	266.56	J/mol×K	652.03	Joback Method
cpg	254.91	J/mol×K	613.20	Joback Method
cpg	242.46	J/mol×K	574.38	Joback Method
cpg	229.17	J/mol×K	535.56	Joback Method
cpg	215.02	J/mol×K	496.74	Joback Method
cpg	199.99	J/mol×K	457.91	Joback Method
cpl	192.63	J/mol×K	298.15	NIST Webbook
hfust	10.00	kJ/mol	189.64	NIST Webbook
hfust	10.00	kJ/mol	189.60	NIST Webbook
hfust	10.00	kJ/mol	189.60	NIST Webbook
hvapt	41.20	kJ/mol	415.50	NIST Webbook
hvapt	37.06	kJ/mol	432.00	NIST Webbook
rho1	964.63	kg/m <sup>3</sup>	293.10	KDB
sfust	52.73	J/mol×K	189.64	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.38045e+01
Coeff. B	-3.38746e+03
Coeff. C	-6.34450e+01
Temperature range (K), min.	314.05
Temperature range (K), max.	462.30

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.04880e+01
Coeff. B	-7.97423e+03
Coeff. C	-9.62230e+00
Coeff. D	5.26186e-06
Temperature range (K), min.	355.15
Temperature range (K), max.	476.15

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1569693&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1569693&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1886">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1886</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemo.com/doc/models/crippen_log10ws">https://www.chemo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol1886.mol">https://www.thermo.com/files/research/kdb/mol/mol1886.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>af:</b>	Acentric Factor
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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