

# Diethylmalonic acid, heptyl 4-methoxyphenyl ester

Inchi:	InChI=1S/C21H32O5/c1-5-8-9-10-11-16-25-19(22)21(6-2,7-3)20(23)26-18-14-12-17(24-4
InchiKey:	JWDVHQBURYJSAQ-UHFFFAOYSA-N
Formula:	C21H32O5
SMILES:	CCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccc(OC)cc1
Mol. weight [g/mol]:	364.48

## Physical Properties

Property code	Value	Unit	Source
gf	-341.28	kJ/mol	Joback Method
hf	-882.28	kJ/mol	Joback Method
hfus	43.15	kJ/mol	Joback Method
hvap	84.70	kJ/mol	Joback Method
log10ws	-5.55		Crippen Method
logp	4.921		Crippen Method
mcvol	303.740	ml/mol	McGowan Method
pc	1253.92	kPa	Joback Method
rinpola	2461.00		NIST Webbook
tb	883.31	K	Joback Method
tc	1090.05	K	Joback Method
tf	534.34	K	Joback Method
vc	1.159	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	967.36	J/molxK	883.31	Joback Method
cpg	983.26	J/molxK	917.77	Joback Method
cpg	997.90	J/molxK	952.22	Joback Method
cpg	1011.31	J/molxK	986.68	Joback Method
cpg	1023.53	J/molxK	1021.13	Joback Method
cpg	1034.59	J/molxK	1055.59	Joback Method
cpg	1044.52	J/molxK	1090.05	Joback Method
dvisc	0.0003350	Paxs	534.34	Joback Method
dvisc	0.0001804	Paxs	592.50	Joback Method

dvisc	0.0001085	Paxs	650.66	Joback Method
dvisc	0.0000710	Paxs	708.82	Joback Method
dvisc	0.0000495	Paxs	766.99	Joback Method
dvisc	0.0000363	Paxs	825.15	Joback Method
dvisc	0.0000277	Paxs	883.31	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369842&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369842&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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