

# 2-epi-Ziza-6(13)-en-3-one

<b>Other names:</b>	2-epi-Ziza-6(13)-en-3-one (methyl group in 2-«alpha»-position)
<b>Inchi:</b>	InChI=1S/C15H22O/c1-9-12-7-13(16)10(2)15(12)6-5-11(8-15)14(9,3)4/h10-12H,1,5-8H2
<b>InchiKey:</b>	RVQOGXCBZWSGSS-SBXIWPSPSA-N
<b>Formula:</b>	C15H22O
<b>SMILES:</b>	<chem>C=C1C2CC(=O)C(C)C23CCC(C3)C1(C)C</chem>
<b>Mol. weight [g/mol]:</b>	218.33

## Physical Properties

Property code	Value	Unit	Source
gf	137.56	kJ/mol	Joback Method
hf	-210.51	kJ/mol	Joback Method
hfus	12.71	kJ/mol	Joback Method
hvap	50.55	kJ/mol	Joback Method
log10ws	-3.71		Crippen Method
logp	3.594		Crippen Method
mcvol	186.900	ml/mol	McGowan Method
pc	2183.60	kPa	Joback Method
rinpol	1657.00		NIST Webbook
rinpol	1660.00		NIST Webbook
ripol	2295.00		NIST Webbook
ripol	2295.00		NIST Webbook
tb	629.48	K	Joback Method
tc	866.95	K	Joback Method
tf	426.81	K	Joback Method
vc	0.716	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	541.93	J/molxK	629.48	Joback Method
cpg	563.88	J/molxK	669.06	Joback Method
cpg	584.68	J/molxK	708.64	Joback Method
cpg	604.63	J/molxK	748.22	Joback Method
cpg	624.03	J/molxK	787.79	Joback Method

cpg	643.18	J/mol×K	827.37	Joback Method
cpg	662.38	J/mol×K	866.95	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R198737&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R198737&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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