

# cis-11-Eicosenoic acid

<b>Other names:</b>	(11Z)-eicosenoic acid (Z)-11-Eicosenic acid (Z)-icos-11-enoic acid cis-11-eicosenoic acid (gondoic acid)
<b>Inchi:</b>	InChI=1S/C20H38O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20(21)22/h9-10
<b>InchiKey:</b>	BITHHVYVYSMSWAG-KTKRTIGZSA-N
<b>Formula:</b>	C20H38O2
<b>SMILES:</b>	CCCCCCCCC=CCCCCCCCCCC(=O)O
<b>Mol. weight [g/mol]:</b>	310.51
<b>CAS:</b>	5561-99-9

## Physical Properties

Property code	Value	Unit	Source
gf	-68.00	kJ/mol	Joback Method
hf	-603.72	kJ/mol	Joback Method
hfus	53.45	kJ/mol	Joback Method
hvap	83.50	kJ/mol	Joback Method
log10ws	-7.15		Crippen Method
logp	6.889		Crippen Method
mcvol	295.800	ml/mol	McGowan Method
pc	1164.04	kPa	Joback Method
rinpol	2356.90		NIST Webbook
rinpol	2356.90		NIST Webbook
tb	807.21	K	Joback Method
tc	989.21	K	Joback Method
tf	420.83	K	Joback Method
vc	1.161	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	929.92	J/mol×K	807.21	Joback Method
cpg	947.52	J/mol×K	837.54	Joback Method
cpg	964.26	J/mol×K	867.88	Joback Method

cpg	980.19	J/molxK	898.21	Joback Method
cpg	995.34	J/molxK	928.54	Joback Method
cpg	1009.77	J/molxK	958.87	Joback Method
cpg	1023.52	J/molxK	989.21	Joback Method
dvisc	0.0004374	Paxs	485.23	Joback Method
dvisc	0.0016119	Paxs	420.83	Joback Method
dvisc	0.0001611	Paxs	549.62	Joback Method
dvisc	0.0000732	Paxs	614.02	Joback Method
dvisc	0.0000386	Paxs	678.42	Joback Method
dvisc	0.0000228	Paxs	742.81	Joback Method
dvisc	0.0000146	Paxs	807.21	Joback Method
hfust	49.70	kJ/mol	296.50	NIST Webbook
hvapt	143.50	kJ/mol	298.15	Vaporization, Sublimation and Fusion Enthalpies of Some Saturated and Unsaturated Long Chain Fatty Acids by Correlation Gas Chromatography

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.57636e+01
Coeff. B	-6.23178e+03
Coeff. C	-1.33846e+02
Temperature range (K), min.	536.52
Temperature range (K), max.	730.07

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5561999&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5561999&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Vaporization, Sublimation and Fusion Enthalpies of Some Saturated and Unsaturated Long Chain Fatty Acids by Correlation Gas Chromatography:** <https://www.doi.org/10.1021/je5009729>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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