

# Isoborneol

**Other names:**

Bicyclo[2.2.1]heptan-2-ol, 1,7,7-trimethyl-, exo-  
exo-2-Hydroxy-1,7,7-trimethylnorbornane  
Isobornyl alcohol  
exo-1,7,7-Trimethylbicyclo(2.2.1)heptan-2-ol  
exo-2-Bornanol  
exo-2-Camphanol  
DL-Isoborneol  
Exoborneol  
Isoborneol, DL-  
Isocamphol  
exo-1,7,7-Trimethylbicyclo(2.2.1)-2-heptanol  
Borneol, exo-  
2-exo-Bornyl alcohol  
Bicyclo(2.2.1)heptan-2-ol, 1,7,7-trimethyl-, (1R,2R,4R)-rel-  
NSC 26350

**Inchi:**

InChI=1S/C10H18O/c1-9(2)7-4-5-10(9,3)8(11)6-7/h7-8,11H,4-6H2,1-3H3/t7?,8-,10?/m1/s

**InchiKey:**

DTGKSKDOIYIVQL-CCNFQMFYSA-N

**Formula:**

C<sub>10</sub>H<sub>18</sub>O

**SMILES:**

CC1(C)C2CCC1(C)C(O)C2

**Mol. weight [g/mol]:**

154.25

**CAS:**

124-76-5

## Physical Properties

Property code	Value	Unit	Source
gf	-20.50	kJ/mol	Joback Method
hf	-272.72	kJ/mol	Joback Method
hfus	9.46	kJ/mol	Joback Method
hvap	51.61	kJ/mol	Joback Method
log10ws	-2.45		Crippen Method
logp	2.194		Crippen Method
mcvol	135.910	ml/mol	McGowan Method
pc	3166.83	kPa	Joback Method
rinpol	1146.00		NIST Webbook
rinpol	1144.00		NIST Webbook
rinpol	1159.00		NIST Webbook
rinpol	1155.00		NIST Webbook
rinpol	1149.00		NIST Webbook

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ripol	1642.00		NIST Webbook
ripol	1665.00		NIST Webbook
ripol	1664.00		NIST Webbook
ripol	1642.00		NIST Webbook
ripol	1660.00		NIST Webbook
tb	529.27	K	Joback Method
tc	730.21	K	Joback Method
tf	334.96	K	Joback Method
vc	0.514	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	354.30	J/mol×K	529.27	Joback Method
cpg	370.25	J/mol×K	562.76	Joback Method
cpg	385.05	J/mol×K	596.25	Joback Method
cpg	398.92	J/mol×K	629.74	Joback Method
cpg	412.02	J/mol×K	663.23	Joback Method
cpg	424.55	J/mol×K	696.72	Joback Method
cpg	436.70	J/mol×K	730.21	Joback Method
hsubt	41.10	kJ/mol	415.00	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C124765&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C124765&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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