

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U351943&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/61-804-1/2-2-2-2-2-2-2-Trifluoroacetyl-oxyethoxy-ethoxy-ethyl-2-2-2-trifluoroace>

Generated by Cheméo on 2024-04-24 10:34:01.434831673 +0000 UTC m=+16244090.355408984.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.