

# Formic acid, 2,6-dimethoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C9H10O4/c1-11-7-4-3-5-8(12-2)9(7)13-6-10/h3-6H,1-2H3
<b>InchiKey:</b>	JVJVENQWWSMNNJ-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O4
<b>SMILES:</b>	COc1cccc(OC)c1OC=O
<b>Mol. weight [g/mol]:</b>	182.17

## Physical Properties

Property code	Value	Unit	Source
gf	-296.47	kJ/mol	Joback Method
hf	-497.74	kJ/mol	Joback Method
hfus	18.18	kJ/mol	Joback Method
hvap	53.18	kJ/mol	Joback Method
log10ws	-1.61		Crippen Method
logp	1.239		Crippen Method
mcvol	133.090	ml/mol	McGowan Method
pc	3235.66	kPa	Joback Method
rinqol	1460.00		NIST Webbook
tb	557.88	K	Joback Method
tc	766.63	K	Joback Method
tf	351.34	K	Joback Method
vc	0.502	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	302.41	J/molxK	557.88	Joback Method
cpg	313.73	J/molxK	592.67	Joback Method
cpg	324.58	J/molxK	627.46	Joback Method
cpg	334.94	J/molxK	662.26	Joback Method
cpg	344.79	J/molxK	697.05	Joback Method
cpg	354.11	J/molxK	731.84	Joback Method
cpg	362.87	J/molxK	766.63	Joback Method
dvisc	0.0009244	Paxs	351.34	Joback Method
dvisc	0.0006083	Paxs	385.76	Joback Method

dvisc	0.0004287	Paxs	420.19	Joback Method
dvisc	0.0003186	Paxs	454.61	Joback Method
dvisc	0.0002468	Paxs	489.03	Joback Method
dvisc	0.0001978	Paxs	523.46	Joback Method
dvisc	0.0001629	Paxs	557.88	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U368910&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U368910&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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