

# Isothiocyanic acid, (p-phenylazo)phenyl ester

<b>Other names:</b>	p-Phenylazophenyl isothiocyanate Diazene, (4-isothiocyanatophenyl)phenyl-
<b>Inchi:</b>	InChI=1S/C13H9N3S/c17-10-14-11-6-8-13(9-7-11)16-15-12-4-2-1-3-5-12/h1-9H
<b>InchiKey:</b>	ZTXNMMXJFVCQPD-UHFFFAOYSA-N
<b>Formula:</b>	C13H9N3S
<b>SMILES:</b>	S=C=Nc1ccc(N=Nc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	239.30
<b>CAS:</b>	7612-96-6

## Physical Properties

Property code	Value	Unit	Source
hf	481.23	kJ/mol	Joback Method
hvap	66.86	kJ/mol	Joback Method
log10ws	-4.43		Crippen Method
logp	4.836		Crippen Method
mcvol	179.900	ml/mol	McGowan Method
pc	2358.78	kPa	Joback Method
tb	850.33	K	Joback Method
tc	1147.09	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7612966&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7612966&amp;Units=SI</a>

## Legend

hf: Enthalpy of formation at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature

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